

The 1<sup>st</sup> Global Inorganic Discussion Weekday Virtual Poster Competition

#gidw2020

On Twitter, July 9<sup>th</sup> and 10<sup>th</sup>, 2020





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### A note from the organizers...

We welcome you to the inaugural #gidw2020 virtual poster event – a 48 h online symposium hosted on Twitter. This poster event was spurred on by the success of our virtual lecture series (Global Inorganic Discussion Weekdays or GIDW, for short), which brings the inorganic (organometallic, main-group, materials, etc.) community together online! The name, GIDW is an homage to the popular regional conferences known as Inorganic Discussion Weekends that occur routinely in the US and Canada (and around the world with other names).

Given the current social distancing policies and travel restrictions, many of our community's standard conferences have been cancelled, preventing us from sharing our progress in science. To overcome scientific isolation and to continue celebrating our advances in inorganic chemistry, we have now organized 15 weeks of online seminars, comprising 56 speakers from Canada, USA, UK, Taiwan, Germany, France, Belgium, Germany, Israel, to name a few.

For this poster event, we would like to thank our sponsors: CIC, CSC-ID, The Wilkinson Charitable Trust, Dalton Transactions and deGruyter publishing for prizes as well as our panel of judges who will be engaging with our presenters over the next 48 h. We also thank all of our participants for submitting an abstract and the inorganic community for their continued support.

Best wishes,



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Saurabh Chitnis

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### **Poster List**

#	Name	Affiliation	Title
P1	Stephen Aderinto	University of Sheffield	New self-assembled hetero-metallomacrocycles
			as biomolecular probes and therapeutics
P2	Jeanette Adjei	Western University	Exploring New Reaction Partners Towards the
<b>D</b> 2			Activation Phosphine Gas (PH <sub>3(g)</sub> )
P3	Mashael Alharbi	Cardiff University	Synthesis, characterisation and properties of tris(3,5-difluorophenyl)borane
P4	Sk Amanullah	Indian Association for the	The role of porphyrin peripheral substituents in
		Cultivation of Science	determining the reactivities of ferrous nitrosyl
			species
P5	Kori Andrea	Memorial University	Iron Amino-phenolate Complexes for CO <sub>2</sub>
			Activation: Selectivity Control in Ligand Design
P6	Nayanthara Asok	York University	Unexpected Reactivity of Structurally Diverse,
			Neutral, Hypervalent P Compounds
P7	Collen Bapiro	Midlands State University	Photocatalytic degradation of 2,4 Dichlorophenol
			by the Nickel Zinc ferrite graphene oxide nanocomposite
P8	Zaira Barrera	University of California, Irvine	Electrochemistry and NMR Experiments of
	Zana Barrera		Reactions Containing Nickel(II) Complexes and
			$CO_2$ in Methanol
P9	Deepika Bedi	Texas Tech University	Polynuclear Lanthanide-Diketonato Clusters:
			Catalytic Hydroboration of Carboxamides and
			Esters
P10	Roman Belli	University of Minnesota	Umpolung of P-H Bond Activation for
P11	Ali Ben Saida	Liniversity of Nomur	Hydrophosphination
PII	All Ben Salda	University of Namur	Pushing the Lewis Acidity Boundaries of Boron Compounds with Non-Planar Triarylboranes
			Derived from Triptycenes
P12	Jordan Bentley	York University	Fluorescent Lewis Adducts: A Practical Guide to
		,	Relative Lewis Acidity
P13	Moulika Bhattacharyya	University of Windsor	Triple Decker Sandwich Complexes Containing
			Six Membered Ring
P14	Milan Bisai	CSIR-NCL Pune	Synthesis and Reactivity of a Hypersilylsilylene
P15	Hannah Bloomfield	University of Winnipeg	Investigating coordination tendencies of low
			valent Ti(III) and V(II) (Abstract Retracted)
P16	Evan Bonnand	Cal State Polytechnic	Synthesis of iron nitrosyl complexes with
D47	Andrui Porus	University, Pomona	diethyldithiocarbamate ligands
P17	Andryj Borys	University of Edinburgh	The Phospha-Bora-Wittig Reaction
P18	Sarah Boudreau	Cape Breton University	Birch Bark-Oil (Maskwiomin): Synthesis and
			Analysis of an L'nu Indigenous Skin Ointment



D10	Jamoy Bower	The Obio State University	Now Chamistry of Copper Helidae: Alkyl C H
P19	Jamey Bower	The Ohio State University	New Chemistry of Copper Halides: Alkyl C-H Fluorination and N/O/S-Nitrosation
P20	David Brossault	University of Cambridge	Self-assembly of nanoparticles in emulsion: a green approach to produce functional silica composites
P21	Pavan C	Acharya and BM Reddy College of Pharmacy	Nanocomposites and their recent advancements
P22	Rachele Carafa	Ryerson University	A DFT and Experimental Investigation of the Ring Stability of Thionylphosphazenes
P23	Jamie Carden	Cardiff University	Unlocking the catalytic potential of tris(3,4,5- trifluorophenyl)borane using microwave irradiation
P24	Kurtis Carsch	Harvard University	Ligand Field Inversion in Sterically Confined Copper Architectures
P25	Cole Carter	Northwestern University	Bis-Ferrocenyl-Pyridinediimine Trinuclear Mixed- Valent Complexes with Metal-Binding Dependent Electronic Coupling
P26	Miguel Chacon Teran	Universidad Simon Bolivar (VE) / Universität Regensburg (DE)	One ligand, two approach.
P27	Yukun Cheng	University of Minnesota	Synthesis of Pentasubstituted 2-Aryl Pyrroles from Boryl and Stannyl Alkynes via One-Pot Sequential Ti-Catalyzed [2+2+1] Pyrrole Synthesis/Cross Coupling Reactions
P28	Saurabh Chitnis	Dalhousie University	Hydrostibination of acetylenes: a mechanism
P29	Marissa Clapson	University of Calgary	Cobalt PCP Carbene Complexes for the Activation of Small Molecules
P30	Omar Coughlin	Nottingham Trent University	Diverse Structure and Reactivity of Pentamethylcyclopentadienyl Antimony(III) Cations
P31	Gloria D'Amaral	Ryerson University	Synthesis and Characterization of Flexible and Rigid Stannane Intermediates & Tin Polymers
P32	Cody Daneluik	Brock University	Mononuclear and pseudo-dinuclear Dy(III) single-molecule magnets in an "ON-OFF" state
P33	Pargol Daneshmandkashani	University of British Columbia	Hydroaminoalkylation with aliphatic amines enabled by the mecha-nistic investigation of cyclic N,O-chelating ureate tantalum catalysts
P34	Johanna de Jong	Western University	Towards the Preparation of Azide-Modified Silver Chalcogenolate Nanoclusters
P35	Carlos Diaz	University of British Columbia	Structure-Activity Relationships in Cationic and Neutral Indium Complexes for Ring-Opening Polymerization Reactions
P36	Hai Dong	University of Michigan	The Fe <sub>2</sub> (NO) <sub>2</sub> Diamond Core: A Unique Structural Motif In Non-Heme Iron-NO Chemistry



P37	Liam Donnelly	University of Edinburgh	Synthesis and Structures of Anionic Rhenium
_			Polyhydride Complexes of Boron-Hydride
			Ligands and their Application in Catalysis
P38	Lorna Doyle	Trinity College Dublin	Activation of a Mn(II)Mn(III)-Peroxide with
			Relevance to the Catalytic Cycle of Class Ib
			RNRs
P39	Luis Duarte	University of Campinas	All-solution-processed organic light-emitting
			diodes based on a novel and versatile
			benzothiazole-salophen derivative
P40	Maeve Dufour	University of Windsor	Octaboraneyl Complexes of Nickel: Monomers
			for Redox-Active Coordination Polymers
P41	Gabriel Durin	French Alternative Energies	Efficient Catalytic Hydrogenolysis of Halosilanes
		and Atomic Energy	to Hydrosilanes Using an Iridium Pincer
		Commission	Complex and Superbases
P42	Matthew Elsby	University of Ottawa	A Direct Comparison of Thiolate vs. Amido
			Donors for Bifunctional Catalysis with Mn and
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P43	Blaine Fiss	McGill University	Solvent-free mechanochemical synthesis of
			ultrasmall nickel phosphide nanoparticles and
			their use as catalysts for the hydrogen evolution reaction
P44	Amanda Fogh	University of British Columbia	Vanadium Aminopyridinate Catalysts for the
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P45	Angelo Frei	University of Queensland	Metal Complexes as a Promising Source for
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P46	Connor Frye	University of Minnesota	Ti-Mediated Intermolecular Diamination of
			Alkynes via [4+2]-Cycloaddition Reactions with
			Trapped [2+2] Intermediates
P47	Mathieu Gama	University of Namur	Triptycenyl phosphines as platform for the
		,	design of frustrated Lewis pair
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	Ū.		using Methanol as Liquid Hydrogen Organic
			Carrier
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			trifluorophenyl)borane using microwave
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P50	Camilla Golec	Trent University	The Metalation of Neuronal Peptides with
			Cu(II)/(I) Modulates the Formation of Reactive
			Oxygen Species
P51	Renée Goodman	Universoty of Windsor	Rational Design Approach for Improved
			Thermomechanical Properties in DPP-based
			Polymers
P52	Chatura Goonesinghe	University of British Columbia	Introducing Hemilability to Cationic Indium
			Complexes



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	Matt Gray	The Ohio State University	High-Efficiency Blue Photoluminescence in the Cs <sub>2</sub> NaInCl <sub>6</sub> :Sb <sup>3+</sup> Double Perovskite Phosphor
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P55	Marcus Grocott	University of Birmingham	It's Better with Salt: Aqueous Ring-Opening Metathesis Polymerization at Neutral pH
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		-	Single Step Synthesis via Hydroaminoalkylation
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		Valley	and deficient Cu-salen complexes
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P153	Adrian Torres	Cal State Polytechnic University, Pomona	Trans oxidative addition of arylfluorides to nickel N-heterocyclic carbene complexes
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		Valley	Complexes as C-H Bond Activation Reagents
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			studies of multinuclear benzimidazole-based
			cyclometallated and cationic ruthenium(II)
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		University, Pomona	Bidentate N-heterocyclic Carbene Ligands
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		Islamabad	Biodiesel Formation: Synthesis, Structural
			Elucidation and Computational Studies

\*\* For ease in creation of this booklet, co-authors have been omitted from abstracts.



### P1. New self-assembled hetero-metallomacrocycles as biomolecular probes and therapeutics

<u>Stephen Aderinto</u> University of Sheffield, Sheffield, United Kingdom Graduate Student **Advisor(s):** Prof. Jim A Thomas

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The ability of self-assembled, kinetically-inert, water-soluble homo- and heterometallomacrocycles constructed from kinetically-inert d<sup>6</sup>-cations and possessing highlystructured binding pockets made up of hydrophobic aromatic residues to bind DNA in biologically-relevant conditions has been explored, leading to the synthesis of Complex 1, the first-ever reported tetranuclear Ru(II)<sub>2</sub>Re(I)<sub>2</sub>, heteroleptic polypyridyl complex, which functions as a DNA-targeting sensitizer for PDT.

1. (a) Fujita, M.; Tominaga, M., Hori A., Therrien, B. *Acc. Chem. Res.* **2005**, 38, 369. (b) Thomas, J. A. *Chem. Soc. Rev.* **2007**, 36, 856. (c) Chakrabarty, R.; Mukherjee, P. S.; Stang, P. J. *Chem. Rev.* **2011**, 111(11), 6810. (e) Thomas, J. A. *Dalton Trans.* **2011**, 40, 1**2005**. 2. (a) de Wolf, P.; Heath, S. L.; Thomas, J. A. *Chem. Commun.* **2002**, 2540. (b) de Wolf, P.; Waywell, P.; Hanson, M.; Heath, S. L.; Meijer, A. J. H. M.; Teat, S. J.; Thomas, J. A. *Chem. Eur. J.* **2006**, 12, 2188. 3. (a) Ghosh, D.; Ahmad, H.; Thomas, J. A. *Chem. Commun.* **2009**, 2947. (b) Ahmad H.; Hazel, B. W.; Meijer, A. J. H. M.; Thomas, J. A.; Wilkinson, K. A. *Chem. Eur. J.* **2013**, 19, 5081. (c) Ahmad, H.; Ghosh, D.; Thomas, J. A. *Chem. Commun.* **2014**, 50, 3859. 4. Walker, M. G.; Jarman, P. J.; Gill, M. R.; Tian, X.; Ahmad, H.; Reddy, P. A. N.; McKenzie, L.; Weinstein, J. A.; Meijer, A. J. H. M.; Battaglia, G.; Smythe, C. G. W.; Thomas, J. A. *Chem. Eur. J.* **2016**, 22(17), 5996. 5. Allison, R. R.; Sibata, C. H. *Photodiagn. Photodyn. Ther.* **2010**, 7(2), 61.



# P2. Exploring New Reaction Partners Towards the Activation Phosphine Gas $(PH_3 (g))$

<u>Jeanette Adjei</u> Western University, London, Canada Graduate Student **Advisor(s):** Paul J. Ragogna & Michael A. Kerr

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This project aims to expand the scope of  $PH_3$  chemistry towards the synthesis of new and valuable organophosphorus compounds for use in catalysis and metal coordination. Hydrophosphination reactions were used to explore the ringing opening of vinylcycloprapane and the P-H bond addition of commercially available natural products. Primary and secondary phosphines were initially used to develop the methodology of these reactions before exploring the activation of PH<sub>3</sub>.

1. Svara, J; Weferling, N; Hofmann, T. In Ullman's Encyclopedia of Industrial Chemistry; Wiley-VCH: Weinhem, **2012**; Vol. 27, pp 19-49.

2. Guterman, R.; Gillies, E. R.; Ragogna, P. J. Dalton Trans., 2015, 44, 15664.

3. Beland, V. A.; Ragogna, P. J. ACS Appl. Mater. Interfaces, 2020, In Press.

4. Irwin, L. Cl; Renwick, C. R.; Kerr, M. A. J.Org. Chem., 2018, 83, 6235.

5. Robertson, A.; Bradaric, C.; Frampton, C. S; McNulty, J.; Capretta, Al., *J. Org. Chem.* **2001**, 42, 2609.



#### P3. Synthesis, characterisation and properties of tris(3,5difluorophenyl)borane

<u>Mashael Alharbi</u> Cardiff University, Cardiff, United Kingdom Graduate Student **Advisor(s):** Dr. Rebecca Melen

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Due to extensive applications of Lewis acid boranes such as  $B(C_6F_5)_3$ , this work intends to synthesize novel Lewis acidic compounds.<sup>1</sup> These will be explored in a wide range of reactions, including small molecule activation and frustrated Lewis pair catalysis. Tris(3,5-difluorophenyl)borane has been synthesized and characterized by multinuclear NMR, mass spectroscopy, and XRD. The Lewis acidity has been determined using both the Gutmann Beckett method and the Childs' method.

1- Beckett, M. A.; Brassington, D. S.; Coles, S. J.; Hursthouse, M. B. *Inorg. Chem. Comm.* **2000**, 3, 530-533.

2- Jacobsen, H.; Berke, H.; Doring, S.; Kehr, G.; Erker, G.; Frohlich, R.; Meyer, O. *Organometallics*, **1999**, 18, 1724-1735.

3- Chase, P. A.; Welch, G. C.; Jurca, T.; Stephan, D. W. Angew. Chem., Int. Ed. 2007, 46, 8050-8053.

4- Ménard, G.; Stephan, D. W. J. Am. Chem. Soc. 2010, 132, 1796-1797.



### P4. The role of porphyrin peripheral substituents in determining the reactivities of ferrous nitrosyl species

<u>Sk Amanullah</u> Indian Association for the Cultivation of Science, Kolkata, India Graduate Student **Advisor(s):** Prof. Abhishek Dey

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{FeNO}<sub>7</sub> is an intermediate common to the catalytic cycles of Cd1NiR (nitrite to NO) and CcNiR (nitrite to NH<sub>4</sub><sup>+</sup>), heme-based NiRs, and its reactivity varies dramatically in these enzymes.<sup>[1,2]</sup> A series of iron porphyrinoids, designed to model the electron-withdrawing peripheral substitution as well as the saturation present in heme d<sup>1</sup> in Cd1NiR, and their NO adducts were synthesized and their properties were investigated. The detailed analysis explains the difference between PCET vs NO release.<sup>[3]</sup>

1. Averill, B. A. Chem. Rev. 1996, 96 (7), 2951-2964

- 2. Maia, L. B.; Moura, J. J. G. Chem. Rev. 2014, 114 (10), 5273-5357
- 3. Amanullah, S.; Dey, A. Chem. Sci. 2020, 11, 5909



## P5. Iron Amino-phenolate Complexes for CO<sub>2</sub> Activation: Selectivity Control in Ligand Design

<u>Kori Andrea</u> Memorial University, St John's, Canada Graduate Student **Advisor(s):** Dr Francesca Kerton

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We have recently reported a collection of iron (III) amino-phenolate complexes active towards the selective production of either cyclic or polycarbonates from epoxides and CO<sub>2</sub>. Selectivity towards either product was found to be dependent both the nature of the phenolate donor, electronics and coordination geometries. During coupling reactions, those complexes in a square pyramidal geometry undergo an epoxide deoxygenation step leading to the corresponding bridged-oxo species.

 Andrea, K. A.; Brown, T. R.; Murphy, J. N.; Jagota, D.; McKearney, D.; Kozak, C. M.; Kerton, F. M. *Inorg. Chem.*, **2018**, 57, 13494-13504.
 Andrea, K. A.; Butler, E. D.; Brown, T. R.; Anderson, T. S.; Jagota, D.; Rose, C.; Lee, E. M.; Goulding, S. D.; Murphy, J. N.; Kerton, F. M.; Kozak, C. M.; *Inorg. Chem.*, **2019**, 58, 11231-11240.



#### P6. Unexpected Reactivity of Structurally Diverse, Neutral, Hypervalent P Compounds

<u>Nayanthara Asok</u> York University, Toronto, Canada Graduate Student **Advisor(s):** Dr. Thomas Baumgartner

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Dithienophospholes provide an interesting conjugated framework towards the synthesis of neutral pentavalent phosphorus compounds with FLP characteristics. These hypervalent compounds acquire a square pyramidal structure that is highly unusual for pentavalent phosphorus species. This peculiar geometry imparts interesting chemical properties that will be highlighted in this contribution.

1. Baumgartner, T. Insights on the Design and Electron-Acceptor Properties of Conjugated Organophosphorus Materials. *Acc. Chem. Res.* **2014**, 47, 1613.

2. Green, M. L. H.; Parkin, G. The classification and representation of main group element compounds that feature three-center four-electron interactions. *Dalton Trans.* **2016**, 45, 18784.

3. Osman, F.H.; El-Samahy, F. A. Reactions of  $\alpha$ -Diketones and o-Quinones with Phosphorus Compounds. *Chem. Rev.* **2002**, 102, 629.

4. Baumgartner, T.; Neumann, T.; Wirges. B. The Dithieno[3,2-b:2',3'-d]phosphole System: A Novel Building Block for Highly Luminescent  $\pi$ -Conjugated Materials. *Angew. Chem. Int. Ed.* **2004**, 43, 6197.

5. Ren, Y.; Dienes, Y.; Hettel, S.; Parvez, M.; Hoge, B.; Baumgartner, T. Highly Fluorinated Dithieno[3,2-b:2',3'-d]phospholes with Stabilized LUMO Levels. *Organometallics* **2009**, 28, 734.

6. Marczenko, K. M.; Zurakowski, J. A.; Kindervater, M. B.; Jee, S.; Hynes, T.; Roberts, N.; Park, S.; Werner-Zwanziger, U.; Lumsden, M.; Langelaan, D. N.; Chitnis, S. S. Periodicity in Structure, Bonding, and Reactivity for p-Block Complexes of a Geometry Constraining Triamide Ligand. *Chem. Eur. J.* **2019**, 25, 16414.

7. Maskey, R.; Schädler, M.; Legler, C.; Greb, L. Bis(perchlorocatecholato)silane - A Neutral Silicon Lewis Super Acid. Angew .Chem. Int. Ed. **2018**, 57, 1717.

8. Greb. L. Lewis Superacids: Classifications, Candidates, and Applications. *Chem. Eur. J.* **2018**, 24, 17881.



### P7. Photocatalytic degradation of 2,4 Dichlorophenol by the Nickel Zinc ferrite graphene oxide nanocomposite

<u>Collen Bapiro</u> Midlands State University, Harare, Zimbabwe Undergraduate Student **Advisor(s):** 

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The widespread use of chemicals in manufacturing industries and their frequent release into the environment is the major cause of pollution, worldwide. Inorganic chemistry can be used to synthesize compounds that degrade water pollutants.



### P8. Electrochemistry and NMR Experiments of Reactions Containing Nickel(II) Complexes and CO<sub>2</sub> in Methanol

Zaira Barrera University of California, Irvine, Irvine, USA Undergraduate Student Advisor(s): Jenny Y. Yang

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Reducing CO<sub>2</sub> to formate is a way to recycle it into a sustainable fuel. NMR and CV experiments were conducted to observe the reactivity of  $[Ni(dmpe)2][PF6]_2$  & to see if [HNi(dmpe)2][PF6] reacts with CO<sub>2</sub> to produce formate in MeOH. Formate was seen at 8.5 ppm after adding CO<sub>2</sub> to the Ni hydride. Moreover, CV experiments with  $[Ni(dmpe)2][PF6]_2$  under N<sub>2</sub> revealed that at low scan rates, irreversible formation of the Ni hydride occurred. At high scan rates, a return of the Ni(II/0) peak was observed.

1) Miedaner, A.; DuBois, D. L.; Curtis, C. J.; Haltiwanger, R. C. *Organometallics* **1993**, 12 (2), 299-303.

2) Tsay, C.; Livesay, B. N.; Ruelas, S.; Yang, J. Y. *J. Am. Chem. Soc.* **2015**, 137, 44, 14114-14121.



### P9. Polynuclear Lanthanide-Diketonato Clusters: Catalytic Hydroboration of Carboxamides and Esters

Deepika Bedi Texas Tech University, Lubbock, USA Graduate Student Advisor(s): Prof. Michael Findlater

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Catalytic reduction of carboxamides into their corresponding amines is an attractive but extremely challenging transformation. We have shown the direct deoxygenation of carboxamides using earth-abundant lanthanum catalysts in the presence of HBPin, delivers good to excellent yields of amines. The reaction was screened for a myriad of substrates and shows excellent functional group tolerance. Furthermore, this strategy was successfully employed to a broad range of esters.



#### P10. Umpolung of P-H Bond Activation for Hydrophosphination

Roman Belli University of Minnesota, USA Post-doctoral Fellow Advisor(s): Lisa Rosenberg

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Efforts to develop a hydrophosphination catalyst system based on complexes that contain phosphenium ligands is described. A key step in this proposed novel catalysis is activation of a P-H bond via hydride abstraction, which we utilized to prepare phosphenium complexes from their respective secondary phosphine precursors. The viability of the additional steps for this proposed catalysis, electrophilic P-C bond formation and hydride transfer, are demonstrated even for substrates such as ethylene.



#### P11. Pushing the Lewis Acidity Boundaries of Boron Compounds with Non-Planar Triarylboranes Derived from Triptycenes

<u>Ali Ben Saida</u> University of Namur, Namur, Belgium Graduate Student **Advisor(s):** Berionni Guillaume

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Bending the archetypal planar trigonal boron atom environment of triarylboranes results in exceptional enhancement of the boron Lewis acidity without requiring electron-withdrawing groups. The bora-phosphatriptycene, which we found to be one of the strongest Lewis acids known up to date and which showed promising reactivities for activating carbon-halogen bonds.<sup>[1]</sup> This bifunctional Lewis acid-base compounds allow the formation of new polymer material.

1. Ben Saida, A., Chardon, A., Osi, A., Tumanow, N., Wouters, J., Adjieufack, A., Champagne, B., Berionni, G. *Angew. Chem. Int. Ed.*, **2019**, 58, 16889.



#### P12. Fluorescent Lewis Adducts: A Practical Guide to Relative Lewis Acidity

<u>Jordan Bentley</u> York University, Toronto, Canada Graduate Student **Advisor(s):** Christopher Caputo

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Recently, we developed a new fluorescence-based method to evaluate the acidity of a series of Lewis acids across the p- and d-blocks of the periodic table with great precision. We have now considerably expanded the scope of the Fluorescent Lewis Adduct (FLA) method by systematically investigating the strength of over 50 Lewis acids in toluene; some of which have never been experimentally measured before.

(1) Gaffen, J. R.; Bentley, J. N.; Torres, L. C.; Chu, C.; Baumgartner, T.; Caputo, C. B. *Chem.* **2019**, 5, 1567-1583.



#### P13. Triple Decker Sandwich Complexes Containing Six Membered Ring

<u>Moulika Bhattacharyya</u> University of Windsor, Windsor, ON, Canada, Canada Post-doctoral Fellow **Advisor(s):** Sundar Ghosh

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Compounds containing flat ring made of boron are very rare and only a handful of such compounds are known. Recently, we have isolated various triple-decker sandwich complexes containing puckered to flat middle rings composed of boron and non-boron elements. The complexes show 22-24-valence electrons that represent the lowest electron count triple-decker complexes. The key results of this work will be described.

[1] (a) Greenwood, N. N. Earnshaw, A.; Chemistry of the Elements; 2nd ed.; Butterworth-Heinemann: Oxford, **1997**.

[2] (a) Grimes R. N. Carboranes, 2nd ed.; Academic Press: London, **2011** (b) Grimes, R. N. *J. Organomet. Chem.* **2013**, 747, 4.

[3] (a) Thakur, A.; Chakrahari, K. K. V.; Mondal, B.; Ghosh, S. *Inorg. Chem.* **2013**, 52, 2262;
(b) Mondal, B.; Mondal, B.; Pal, K.; Varghese, B.; Ghosh, S. *Chem. Commun.* **2015**, 51, 3828;
(c) Mondal, B.; Bhattacharyya, M.; Varghese, B.; Ghosh, S. *Dalton Trans.* **2016**, 45, 10999.
[4] (a) Lauher, J. W.; Elian, M.; Summerville, R. H.; Hoffmann, R. *J. Am. Chem. Soc.* 1976, 98, 3219; (b) Jemmis, E. D.; Reddy, A. C. *J. Am. Chem. Soc.* **1990**, 112, 722. (c) Jemmis, E. D.; Reddy, A. C. *J. Am. Chem. Soc.* **1990**, 112, 722. (c) Jemmis, E. D.; Reddy, A. C. *J. Soc.* **1988**, 7, 1561.



#### P14. Synthesis and Reactivity of a HypersilyIsilylene

Milan Bisai CSIR-NCL Pune, Pune, India Graduate Student Advisor(s): Dr. Sakya S. Sen

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Stabilization of an amidinatosilylene with a bulky hypersilyl substituent was realized with the preparation of PhC(NtBu)<sub>2</sub>Si{Si(SiMe<sub>3</sub>)<sub>3</sub>} in more than 90% yield. Subsequent to the synthesis of the silylene, we studied its reactivity toward Me<sub>3</sub>NO, S, Se, and Te. Silaoxirane formation was observed when it was treated with acetone. Formation of tetraphosphilosilane was also realized by the reaction of chlorodiphenylphospine.

1. Bisai, M. K.; Swamy, V. S. V. S. N.; Das, T.; Vanka, K.; Gonnade, R. G.; Sen, S. S. *Inorg. Chem.* **2019**, 58, 10536.



#### P15. Investigating coordination tendencies of low valent Ti(III) and V(II)

(Abstract Retracted) <u>Hannah Bloomfield</u> University of Winnipeg, Winnipeg, Canada Undergraduate Student **Advisor(s):** Jamie Ritch

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Low oxidation state transition metals (Ti<sup>3+</sup>/V<sup>2+</sup>) are underexplored compared to more stable oxidation states (Ti<sup>4+</sup>/V<sup>3+</sup>). We theorized that the strong  $\sigma$ -donating and p-accepting ability of NHCs may render them good spectator ligands for these low valent metals. Various NHCs were reacted with VCl<sub>2</sub>(TMEDA)<sub>2</sub> and TiCl<sub>3</sub>(THF)<sub>3</sub> to study coordination tendencies, complemented by DFT calculations. This poster will highlight our experimental and computational efforts to identify ideal NHCs for stabilization.



### P16. Synthesis of iron nitrosyl complexes with diethyldithiocarbamate ligands

Evan Bonnand Cal State Polytechnic University, Pomona, USA Undergraduate Student Advisor(s): Dr. Stieber

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Literature reports the formation of an iron nitrosyl complex first, followed by the coordination of two dtc ligands. This work explores that approach by varying stoichiometric ratios and types of reagents used with the goal of optimizing reaction conditions and establishing an airstable synthetic method. Deviations from literature include using Fe(III) instead of Fe(II) with both silver and sodium dtc salts. IR and NMR spectroscopy were utilized to verify the formation of the complex.

(1) Additional Materials for the Iron Nitrosyl experiment

http://www.chem.ucla.edu/~bacher/CHEM174/hints/Iron\_Nitrosyl.html (accessed Jun 30, **2020**).

(2) Iron Nitrosyl compounds www.chem.ucla.edu/~bacher/CHEM174/Lectures/...Nitrosyl.pptx (accessed Jun 30, **2020**).



#### P17. The Phospha-Bora-Wittig Reaction

<u>Andryj Borys</u> University of Edinburgh, Edinburgh, Scotland Post-doctoral Fellow **Advisor(s):** Dr Michael Cowley

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Transient phosphaborenes, [Mes\*P=B-NR<sub>2</sub>], undergo a [2+2] cycloaddition with aldehydes and ketones to give 1,2,3-phosphaboraoxetanes, which can be converted into phosphaalkenes using sub-stoichiometric N-heterocyclic carbene or AlBr<sub>3</sub>, with loss of [R<sub>2</sub>N-BO]<sub>3</sub>. In contrast, phosphaalkenes are directly prepared from esters and amides, offering a versatile method for transforming carbonyl compounds into phosphaalkenes via the 'phosphabora Wittig' reaction.



### P18. Birch Bark-Oil (Maskwiomin): Synthesis and Analysis of an L'nu Indigenous Skin Ointment

<u>Sarah Boudreau</u> Cape Breton University, Sydney, Canada Undergraduate Student **Advisor(s):** Dr. Matthias Bierenstiel

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Maskwiomin (birch bark-oil) is an almost forgotten L'nu medicine for the topical treatment of skin conditions (psoriasis, eczema, rashes). The L'nu method of producing the oil involves torrefaction in a campfire using a can-over-can process. The Bierenstiel group has been developing a process to control the conditions of this process with an electrical and temperature-controlled reactor. The goal of this project is to modify the conditions of the reactor to obtain better, reproducible results.

1. Flechter, O. B.; et al. *Pharm. Chem. J.* **2005**, 39, 401-404.

2. Jager, S.; et al. *Molecules* **2008**, 13, 3224-3235.

3. Krasutsky, P. A. Nat. Prod. Rep. 2006, 23, 919-942.



#### P19. New Chemistry of Copper Halides: Alkyl C-H Fluorination and N/O/S-Nitrosation

<u>Jamey Bower</u> The Ohio State University, Columbus, United States of America Graduate Student **Advisor(s):** Shiyu Zhang

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New copper halide complexes are presented that perform alkyl C-H fluorination and N/O/Snitrosation enabled by highly covalent metal-ligand interactions. The complexes are best described as copper(II) antiferromagnetically coupled to a ligand-centered radical, which provides key insights into their reactivity. Together, these studies highlight the synergy between multireference electronic structures and ligand-centered radical reactivity of copper complexes with ambiguous oxidation states.

Bower, J. K.; Cypcar, A. D.; Henriquez, B.; Stieber, S. C. E.; Zhang, S. J. Am. Chem. Soc. 2020, 142, 8514.
 Bower, J. K.; Sokolov, A. Yu.; Zhang, S. Angew. Chem. Int. Ed. 2019, 58, 10225.



### P20. Self-assembly of nanoparticles in emulsion: a green approach to produce functional silica composites

David Brossault University of Cambridge, Cambridge, United Kingdom Graduate Student Advisor(s): Prof. Alex F. Routh

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Silica composites are promising materials, used for biological and environmental applications. However, their preparation is often time-consuming and requires toxic chemicals or high temperatures. This poster presents a green approach to produce metal doped silica composites at room temperature via salt-induced assembly of nanoparticles in aqueous droplets.<sup>[1]</sup> Such a method is fast, tuneable and can easily be adapted to produce functional systems (e.g. Recoverable Photocatalytic Microbeads).<sup>[2]</sup>

1. Brossault, D.; Routh, A. J. Colloid Interface Sci. 2020, 562, 381.

2. Brossault, D.; McCoy, T.; Routh, A. Manuscript under review



#### P21. Nanocomposites and their recent advancements

<u>Pavan C.</u> Acharya and BM Reddy College of Pharmacy, Bangalore, India Undergraduate Student **Advisor(s):** Madhavi BLR

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Nanocomposites are versatile in their applications such as anti-corrosive, healing of bones, sensors, environmental protection, packaging, wastewater treatment, and diagnosis of tumors. Adverse effects are not just due to their individual properties of components but influence their morphology, dispersion, and interfacial characteristics. Advancements of nanocomposites are the production of flexible and rigid packaging and nanofillers creating nanocomposite structures with improved properties.



## P22. A DFT and Experimental Investigation of the Ring Stability of Thionylphosphazenes

Rachele Carafa Ryerson University, Toronto, Canada Graduate Student Advisor(s): Daniel Foucher & Andrew McWilliams

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Thionylphosphazenes have been shown to undergo ring-opening polymerization (ROP) at both elevated temperatures in the melt and at ambient temperatures in the presence of Lewis acids, but the mechanism on how this works is not fully understood. This study aims to apply novel DFT calculations to thionylphosphazenes and some of its derivatives to update current literature calculations while also providing more insight into their ring stability and the ROP mechanism.

McWilliams, A. R.; Gates, D. P.; Edwards, M.; Liable-sands, L. M.; Guzei, I.; Rheingold, A. L.; Manners, I.; April, R. V. *J. Am. Chem. Soc.* **2000**, 122 (6), 8848-8855.



### P23. Unlocking the catalytic potential of tris(3,4,5-trifluorophenyl)borane using microwave irradiation

Jamie Carden Cardiff University, Cardiff, United Kingdom Graduate Student Advisor(s): Dr Rebecca Melen

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In this work we explore the catalytic properties of tris(3,4,5-trifluorophenyl)borane towards hydroboration, and the use of microwave irradiation to allow conventionally difficult hydroboration reactions to occur.

Carden J. L, Gierlichs L. J., Wass D. F., Browne D. L., and Melen R. L., *Chem. Comm.*, **2019**, 55, 318.



#### P24. Ligand Field Inversion in Sterically Confined Copper Architectures

<u>Kurtis Carsch</u> Harvard University, Somerville, United States Graduate Student **Advisor(s):** Theodore A. Betley

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The isolation and characterization of formally high-valent Cu(III) complexes permits a rigorous disambiguation of electronic structure. We demonstrate ligand field inversion is general in which a physical 3d<sup>8</sup> electronic configuration is inaccessible and metal-ligand multiple bonding is attenuated. These findings facilitate refined considerations of how the absence of metal-ligand multiple bonds in late transition metal complexes dictate electronic structure and multi-electron catalysis.

Carsch, K.M.; DiMucci, I.M.; Iovan, D.A.; Li, A.; Zheng, S.-L.; Titus, C.J.; Lee, S.J.; Irwin, K.D.; Nordlund, D.; Lancaster, K.M.; Betley, T.A. *Science* **2019**, 365, 1138.
 Carsch, K. M.; Lukens, J. T.; DiMucci, I. M.; Iovan, D. A.; Zheng, S.-L.; Lancaster, K. M.; Betley, T. A. *J. Am. Chem. Soc.***2020**, 142, 2264.
 DiMucci, I.M.; Lukens, J.T.; Chatterjee, S.; Carsch, K.M.; Titus, C.J.; Lee, S.J.; Nordlund,

D.; Betley, T.A.; MacMillan, S.N.; Lancaster, K.M. J. Am. Chem. Soc. 2019, 141, 18508.



# P25. Bis-Ferrocenyl-Pyridinediimine Trinuclear Mixed-Valent Complexes with Metal-Binding Dependent Electronic Coupling

<u>Cole Carter</u> Northwestern University, Evanston, United States Graduate Student **Advisor(s):** Tobin Marks

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A family of ferrocene substituted Pyridinediimine metal dichloride (Fc<sub>2</sub>PDI)MCl<sub>2</sub> complexes were synthesized and electronically characterized. While the parent Fc<sub>2</sub>PDI ligand does not demonstrate electronic coupling between Fc units, metallation with MCl<sub>2</sub> ions promotes electronic coupling and delocalization between metals in their mixed valent state. DFT, crystallography, and spectroelectrochemistry aid in gaining granular detail about structural and electronic changes in mixed valent species.



#### P26. One ligand, two approach

<u>Miguel Chacon Teran</u> Universidad Simon Bolivar (VE) / Universität Regensburg (DE), Caracas, Venezuela Graduate Student **Advisor(s):** Dr. Rafael Rodriguez Lugo / Dr. Robert Wolf

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Recent findings motivate to look forward  $P(OQuin)_3$ -chemistry.<sup>[1]</sup> In these sense, two new different  $P(OQuin)_3$ -approach toward second row-TM are presented. A  $P(OQuin)_3$ -Ru system, is a suitable catalyst for the dehydrogenative homo-/heterocoupling of silanes providing a TOF of 6000 h<sup>-1</sup>. While a  $P(OQuin)_3$ -Rh system provide an excellent regiocontrol over the dearomative hydroboration of Py and Quin to yield 1,2-dehydroheterocycles under mild condition where additive or base are not required.

[1] (a) Rodriguez-Lugo, R. E.; Chacon-Teran, M. A.; De Leon, S.; Vogt, M.; Rosen-thal, A. J.; Landaeta, V. R., *Dalton Trans.*, **2018**, 47, 2061. (b) Chacon-Teran; M. A.; Rodriguez-Lugo, R. E.; Wolf, R.; Landaeta. V. R., Eur. J. *Inorg. Chem.* **2019**, 39-40, 4336.



# P27. Synthesis of Pentasubstituted 2-Aryl Pyrroles from Boryl and Stannyl Alkynes via One-Pot Sequential Ti-Catalyzed [2+2+1] Pyrrole Synthesis/Cross Coupling Reactions

<u>Yukun Cheng</u> University of Minnesota, Minneapolis, United States Graduate Student **Advisor(s):** Ian Tonks

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Pyrroles are commonly found in bioactive small molecule scaffolds, yet the synthesis of highly-substituted pyrrole cores remains challenging. Herein, we report a catalytic synthesis of 2-(9-BBN or SnR<sub>3</sub>) pyrroles via Ti-catalyzed [2+2+1] heterocoupling of heteroatom-substituted alkynes. The resulting 2-boryl substituted pyrroles can be used in Suzuki reactions in a one-pot fashion, resulting in pentasubstituted 2-aryl pyrroles that are inaccessible via previous [2+2+1] heterocoupling strategies.



#### P28. Hydrostibination of acetylenes: a mechanism

<u>Saurabh Chitnis</u> Dalhousie University, Halifax, Canada Academia - Faculty/Staff **Advisor(s):** 

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Boranes have a low-energy vacant p-orbital and a hydridic B-H bond. This unlocks hydroboration. We have reported stibines, R<sub>2</sub>SbH, having a low-energy  $\sigma^*$  LUMO and hydridic Sb-H bonds, which perform hydrostibination.<sup>[1]</sup> This poster describes experiments to discover the mechanism of this reaction.

[1] Marczenko, K. M.; Zurakowski, J. A.; Bamford, K. L.; MacMillan, J. W. M.; Chitnis, S. S. *Angew. Chem. Int. Ed.*, **2019**, 58, 18096.



#### P29. Cobalt PCP Carbene Complexes for the Activation of Small Molecules

<u>Marissa Clapson</u> University of Calgary, Calgary, Canada Graduate Student **Advisor(s):** Dr. Warren E. Piers

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Co(I)-X (X = CI or Br) PCP carbene complexes can be readily synthesized following treatment of the corresponding tetrahedral Co(II)X<sub>2</sub> complexes with NaHBEt<sub>3</sub> and Gomberg's dimer. Treatment of the Co(I)-X complexes with either CsOH or LiCH<sub>2</sub>TMS, results in the formation of the Co(I)-OH and Co(I)-CH<sub>2</sub>TMS complexes, respectively. Co(I)-OH, when placed under vacuum, forms a Co(I)-O-Co(I)  $\mu$ -oxo complex. Exposure of Co(I)-O-Co(I) to CO<sub>2</sub> results in the rapid formation of a bridging carboxylate species.

(1) Smith, J. D.; Chih, E.; Piers, W. E.; Spasyuk, D. M. Polyhedron 2018, 155, 281.

- (2) LaPierre, E. A.; Piers, W. E.; Gendy, C. Organometallics 2018, 37 (20), 3394.
- (3) Sung, S.; Wang, Q.; Kraemer, T.; Young, R. D. Chem. Sci. 2018.

(4) Burford, R. J.; Piers, W. E.; Ess, D. H.; Parvez, M. J. Am. Chem. Soc. 2014, 136 (8), 3256.



# P30. Diverse Structure and Reactivity of Pentamethylcyclopentadienyl Antimony (III) Cations

<u>Omar Coughlin</u> Nottingham Trent University, Nottingham, United Kingdom Graduate Student **Advisor(s):** Sophie Benjamin

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Heavy Group 15 organometallic complexes are emerging as powerful reagents with the potential to perform specific stoichiometric and catalytic transformations.<sup>[1-3]</sup> Harnessing advances in novel synthetic strategies and inspired by recent reports of main group cations with metallocene structures,<sup>[4-5]</sup> we have synthesised a range of pentamethylcyclopentadienyl antimony(III) cations and probed their diverse structure and reactivity.<sup>[6]</sup>

[1] Planas, O.; Wang, F.; Leutzsch, M.; Cornella, J. Science. 2020, 367, 313.
[2] Stephan, D. W.; Erker, G. Angew. Chem., Int. Ed. 2015, 54, 6400.

[3] Pan, B.; Gabbai, F. P. Angew. Chem., Int. Ed. J. Am. Chem. Soc. 2014, 136, 9564.

[4] Zhou, J.; L. Liu, L.; L. Cao, L.; W. Stephan, D. Chem. 2018, 4, 2699.

[5] N. Jones, J.; H. Cowley. A; L. B. Macdonald, C. Chem. Commun. 2002, 1520.

[6] Coughlin, O.; Krämer, T.; Benjamin, S.L. Dalton Trans. 2020, 49, 1726.



#### P31. Synthesis and Characterization of Flexible and Rigid Stannane Intermediates & Tin Polymers

<u>Gloria D'Amaral</u> Ryerson University, Toronto, Canada Graduate Student **Advisor(s):** Dr. Daniel Foucher

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Polystannanes are a unique class of polymers that consist of a backbone of covalently linked tin atoms, however they display an extreme sensitivity to light and moisture. Recently, the Foucher group has demonstrated significant stability improvements when there is an increase in electron density around the tin center via hypercoordination. Using this evidence, my project focuses on synthesizing light and moisture stable polystannanes and exploring their semiconductive properties.

1. Pau, J.; D'Amaral, G. M.; Lough, A. J.; Wylie, R. S.; Foucher, D. A. *Chem. Eur. J.* **2018**, 24, 18762-18771.



# P32. Mononuclear and pseudo-dinuclear Dy(III) single-molecule magnets in an "ON-OFF" state

<u>Cody Daneluik</u> Brock University, Niagara Falls, Canada Undergraduate Student **Advisor(s):** Melanie Pilkington & Theocharis Stamatatos

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A new hexa-nucleating N<sub>3</sub>O<sub>3</sub> organic chelate (L<sub>3</sub>-) was synthesized and used for the preparation of low-coordinate, mononuclear Dy(III) single-molecule magnets (SMMs). Our first attempts led to the crystallization and structural elucidation of the 7-coordinate compound [DyL(H<sub>2</sub>O)], which is singly H-bonded to its neighbouring monomers, yielding isolated pseudodimers that are well-separated from one another. The H-bonded dimers show interesting magnetic dynamics mainly due to exchange-bias phenomena from the communication between the two spin systems. Chemical reactivity studies have allowed us to alter the magnetic properties of these H-bonded species through the synthesis of the non-SMM complex [DyL(THF)]. Details of synthetic aspects and rationale beyond preparation of these complexes, as well as their complete magnetic characterization, will be reported in this presentation.



# P33. Hydroaminoalkylation with aliphatic amines enabled by the mechanistic investigation of cyclic N,O-chelating ureate tantalum catalysts

Pargol Daneshmandkashani University of British Columbia, Vancouver, Canada Post-doctoral Fellow Advisor(s): Prof. Laurel Schafer

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The first optimized N,O-chelated cyclic ureate tantalum catalyst can be used for the hydroaminoalkylation of 1-octene with challenging N-alkyl secondary amine substrates. This catalyst offers turnover frequencies of up to  $60 h^{-1}$  and short reaction times of approximately 1 hour. Kinetic isotope effect (KIE) studies reveal that catalytic turnover is the limiting step. Mechanistic studies show that catalytic turnover is promoted by electron rich amine substrates (Hammett plot).



# P34. Towards the Preparation of Azide-Modified Silver Chalcogenolate Nanoclusters

<u>Johanna de Jong</u> Western University, London, Canada Undergraduate Student **Advisor(s):** Dr. Mark Workentin, Dr. John F. Corrigan

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Ag nanoclusters (NCs) have applications in optical imaging, chemical sensing, and catalysis, but their ability to undergo interfacial surface chemistry is currently limited by the inertness of their organic ligands. Here we discuss progress towards functionalizing a Ag chalcogenolate NC with an azide capable of a biorthogonal click reaction: the Strain-Promoted Alkyne-Azide Cycloaddition. The ability to link molecules to the surface of NCs enables their properties to be tuned and enhanced.



# **35. Structure-Activity Relationships in Cationic and Neutral Indium Complexes for Ring-Opening Polymerization Reactions**

<u>Carlos Diaz</u> University of British Columbia, Vancouver, Canada Graduate Student **Advisor(s):** Parisa Mehrkhodavandi

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Herein, we present different cationic indium with different side-arm donor groups and evaluate their effects on polymerization rates of cyclic ethers. Similarly, neutral complexes are evaluated for the ROP of cyclic esters and copolymerization of epoxides with carbon dioxide. Structure-function studies were conducted to better understand the different mechanisms involved and potentially improve the reactivity and control over the polymerizations



# P36. The Fe<sub>2</sub>(NO)<sub>2</sub> Diamond Core: A Unique Structural Motif In Non-Heme Iron-NO Chemistry

Hai Dong University of Michigan, Ann Arbor, United States Graduate Student Advisor(s): Nicolai Lehnert

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Many hs-{FeNO}<sub>8</sub> complexes disproportionate by forming dinitrosyl iron complexes (DNICs). However, the mechanism of this reaction is not understood. The hs-{FeNO}<sub>7</sub> complex, [Fe(TPA)(NO)(OTf)](OTf) (1) was synthesized and characterized by different spectroscopic methods and X-ray crystallography. Upon reduction of 1 with 1 eq. of CoCp<sub>2</sub> we isolated and structurally characterized a novel dimer, [Fe<sub>2</sub>(TPA)<sub>2</sub>(NO)<sub>2</sub>](OTf)<sub>2</sub>, which we propose to be a potential intermediate of DNIC formation.

Dong, H. T.; Speelman, A. L.; Kozemchak, C. E.; Sil, D.; Krebs, C.; Lehnert, N. *Angew. Chem. Int. Ed.* **2019**, 58, 17695-17699.



# P37. Synthesis and Structures of Anionic Rhenium Polyhydride Complexes of Boron-Hydride Ligands and their Application in Catalysis

<u>Liam Donnelly</u> University of Edinburgh, Edinburgh, United Kingdom Graduate Student **Advisor(s):** Prof. Jason Love and Dr Stephen Thomas

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The rhenium complex, [K(DME)(18-c-6][ReH<sub>7</sub>(BPin)<sub>3</sub>] comprising hydride and boron ligands only, has been synthesized by exhaustive deoxygenation of the perrhenate anion (ReO<sub>4</sub>-) with pinacol borane (HBPin). The structure of this complex was analysed by X-ray crystallography, NMR spectroscopy, and DFT calculations. This complex is a catalyst for the hydroboration of pyridines under simple operating procedures and is also a reagent for the C-H borylation of toluene.

Donnelly, L. J.; Parsons, S.; Morrison, C. A.; Thomas, S. P.; Love, J. B. Submitted



# P38. Activation of a Mn(II)Mn(III)-Peroxide with Relevance to the Catalytic Cycle of Class Ib RNRs

Lorna Doyle Trinity College Dublin, Dublin, Ireland Graduate Student Advisor(s): Aidan R. McDonald

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Ribonucleotide reductase enzymes (RNR) convert ribonucleotides to deoxyribonucleotides.<sup>[1]</sup> The biomimetic complex, [Mn(II)<sub>2</sub>(BPMP)(OAc)<sub>2</sub>](CIO<sub>4</sub>) (1) reacts with superoxide, the RNR oxidising agent, to form an Mn(II)Mn(III)-peroxide which reacts with acids to form an Mn(III)Mn(IV) species. mimicking the cycle of RNR and suggests the role acid activation in the cycle of RNRs.<sup>[2],[3]</sup>

[1] Cotruvo, J. A.; Stubbe, J. Annu. Rev. Biochem. 2011, 80 (1), 733.
 [2] Cotruvo, J. A.; Stich, T. A.; Britt, R. D.; Stubbe, J. J. Am. Chem. Soc. 2013, 135 (10), 4027. [3] (a) Suzuki, M.; Mikuriya, M.; Murata, S.; Uehara, A.; Oshio, H.; Kida, S.; Saito, K. Bull. Chem. Soc. Jpn. 1987, 60 (12), 4305 (b) Blanchard, S.; Blondin, G.; RiviÃ<sup>\*</sup>re, E.; Nierlich, M.; Girerd, J.-J. Inorg. Chem. 2007, 42 (15), 4568.



### P39. All-solution-processed organic light-emitting diodes based on a novel and versatile benzothiazole-salophen derivative

Luis Duarte University of Campinas, Campinas, Brazil Post-doctoral Fellow Advisor(s):

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The N,N'-bis(salicylidene)-(2-(3',4'-diaminophenyl)benzothiazole (BTS), that undergoes Excited State Intramolecular Proton Transfer process (ESIPT) was synthesized and characterized as a strategy to construct fluorescence emission covering the entire visible spectrum. BTS was studied by steady-state and time-resolved spectroscopies. Its ESIPT balance and Zn coordination compound allowed the production of all-solution processed WOLEDs in host-guest composites.



# P40. Octaboraneyl Complexes of Nickel: Monomers for Redox-Active Coordination Polymers

<u>Maeve Dufour</u> University of Windsor, Windsor, Canada Undergraduate Student **Advisor(s):** Marcus Drover

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We establish the preparation, characterization, and reactivity of 1,2-bis(di(3dicyclohexylboraneyl)propylphosphino)-ethane( $P_2B^{Cy_4}$ ). The coordination of  $P_2B^{Cy_4}$  with nickel, results in Ni( $P_2B^{Cy_4}$ )<sub>2</sub> and is reactive with Lewis bases. Addition of 4,4'-bipyridine, creates an air-sensitive coordination polymer. Characterized by solid-state NMR and EPR, a charge transfer polymer was identified from an intramolecular Ni $\rightarrow$ 4,4'-bpy ET, providing oxidized nickel sites and reduced 4,4'-bpy radical anion sites.

Drover, M. W.; Dufour, M.C.; Lesperance-Nantau, L.A.; Noriega, R.P.; Levin, K.; Schurko, R.W. *Chem. -Eur. J.* **2020**, DOI: 10.1002/chem.202001218



# P41. Efficient Catalytic Hydrogenolysis of Halosilanes to Hydrosilanes Using an Iridium Pincer Complex and Superbases

<u>Gabriel Durin</u> French Alternative Energies and Atomic Energy Commission, Paris, France Graduate Student **Advisor(s):** Thibault Cantat

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Hydrosilanes are convenient reductants but are produced via energy intensive processes<sup>3</sup> and generate, after use, silicon oxides wastes such as siloxanes, which are difficult to recycle. These limitations call for the development of catalytic processes able to transform Si-O bonds in siloxanes into Si-H hydrides. Previous reports4 have shown that chlorosilanes can be prepared from siloxanes. We report here an efficient hydrogenolysis of chlorosilanes (up to 79 %) using [(<sup>t</sup>BuPOCOP)IrHCI] complex.

1. Monsigny, L.; Feghali, E.; Berthet, J. C.; Cantat, T. Green Chem., 2018, 20, 1981.

2. Riduan, S. N.; Zhang, Y.; Ying, J. Y. Angew. Chem. Int. Ed., 2009, 48, 3322.

3. Kalchauer, W.; Pachaly, B. In Handbook of Heterogeneous Catalysis.Wiley, **2008**, 2635-2647.

4. (a) Di Giorgio, P. A.; Strong, W. A.; Sommer, L. H.; Whitmore, F. C. *J. Am. Chem. Soc*, **1946**, 68, 1380 (b) Bailey, D. L.; Sommer, L. H.; Whitmore, F. C. *J. Am. Chem. Soc*, **1948**, 70, 435-436 (c) Masaoka, S.; Bannoa, T.; Ishikawa, M. *J. Organomet. Chem.*, **2006**, 691, 174-181 (d) Roberts, J. M.; Eldred, D. V.; Katsoulis, D. E. Ind. Eng. Chem. Res., **2016**, 55, 1813-1818.

5. (a) Tsushima, D.; Igarashi, M.; Sato, K.; Shimada, S. *Chem. Lett.*, **2017**, 46, 1532-1534. See also hydrogenolysis of chlorosilanes in the presence of additives (Nal, NaBAr4<sup>F</sup>) : (b) Beppu, T.; Sakamoto, K.; Nakajima, Y.; Matsumoto, K.; Sato, K.; Shimada, S. *J. Org. Chem.*, **2018**, 869, 75-80. (c) Gluer, A.; Schweizer, J. I.; Karaca, U. S.; Wuertele, C.; Diefenbach, M.; Holthausen, M. C.; Schneider, S. *Inorg. Chem.*, **2018**, 57, 13822-13828.



# P42. A Direct Comparison of Thiolate vs. Amido Donors for Bifunctional Catalysis with Mn and Cu Catalysts

<u>Matthew Elsby</u> University of Ottawa, Ottawa, Canada Graduate Student **Advisor(s):** Prof. R. Tom Baker

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Base-metal complexes that utilize metal-ligand-cooperativity are highly sought. Two new Cu-SNS complexes were synthesized to compare the activity of an amido vs a thiolate donor in outer-sphere carbonyl reactivity mechanistic studies involving a "forced" outer-sphere mechanism led to the design of Mn catalysts. Insight into the Mn oxidation state in the active species in outer-sphere carbonyl reductions employing Mn(II) precatalysts are discussed.

T. Higashi, S. Kusumoto, and K. Nozaki, Cleavage of Si-H, B-H, and C-H Bonds by Metal-Ligand Cooperation, *Chem. Rev.* **2019**, 119, 10393-10402. DOI: 10.1021/acs.chemrev.9b00262



P43. Solvent-free mechanochemical synthesis of ultrasmall nickel phosphide nanoparticles and their use as catalysts for the hydrogen evolution reaction

<u>Blaine Fiss</u> McGill University, Montreal, Canada Graduate Student **Advisor(s):** Dr. Audrey Moores and Dr. Tomislav Friščić

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We demonstrate the mechanochemical application of sodium phosphide as an excellent solidstate phosphorus source for the synthesis of Ni<sub>2</sub>P nanoparticles below 3 nm in diameter, the smallest seen for the mechanochemical synthesis of metal phosphides. These nanoparticles, supported on g-C<sub>3</sub>N<sub>4</sub>, were successful for the hydrogen evolution reaction, generating 233.9 µmol g<sup>-1</sup> h<sup>-1</sup> of hydrogen, over 3 hours using broad spectrum light at room temperature, as well as being readily recyclable and reusable.

Shi, Y.; Zhang, B. *Chem. Soc. Rev.* **2016**, 45, 1529. Fiss, B. G.; Vu, N. N.; Douglas, G.; Do, T. O.; Friščić, T.; Moores, A. *Submitted.* **2020**.



# P44. Vanadium Aminopyridinate Catalysts for the Reductive Coupling of Alcohols

<u>Amanda Fogh</u> University of British Columbia, Vancouver, Canada Undergraduate Student **Advisor(s):** Dr.Laural Schafer

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The reductive coupling of alcohols is beneficial for the re-functionalization of biomass-derived material. Vanadium catalysts were investigated for this transformation. A vanadium aminopyridinate catalyst was able to perform the reductive coupling of alcohols more efficiently than previously reported catalytic systems. Many vanadium aminopyridinate complexes were synthesized towards mechanistic understanding of this catalytic system and provided insight into this underdeveloped field of study.



#### P45. Metal Complexes as a Promising Source for New Antimicrobials?

<u>Angelo Frei</u> University of Queensland, Brisbane, Australia Post-doctoral Fellow **Advisor(s):** Dr. Mark Blaskovich

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Metal complexes are a cornerstone of medicinal chemistry, with compounds such as the anticancer drug Cisplatin still constituting one of the most important chemotherapeutics in the clinics today. However, metals have remained largely unexplored for antimicrobial applications. We have tested the antimicrobial activity of over 300,000 crowd-sourced compounds and found that metal complexes have a significantly higher hit-rate against bacteria and fungi than purely organic molecules.

Frei, A.; Zuegg, J.; Elliott, A. G.; Baker, M.; Braese, S.; Brown, C.; Chen, F.; G. Dowson, C.; Dujardin, G.; Jung, N.; King, A. P.; Mansour, A. M.; Massi, M.; Moat, J.; Mohamed, H. A.; Renfrew, A. K.; Rutledge, P. J.; Sadler, P. J.; Todd, M. H.; Willans, C. E.; Wilson, J. J.; Cooper, M. A.; Blaskovich, M. A. T., *Chem. Sci.* **2020**, 11, 2627.



#### P46. Ti-Mediated Intermolecular Diamination of Alkynes via [4+2]-Cycloaddition Reactions with Trapped [2+2] Intermediates

<u>Connor Frye</u> University of Minnesota, Minneapolis, United States Graduate Student **Advisor(s):** Ian Tonks

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Accessing unsymmetrical  $\alpha$ -diimines through acid-catalyzed condensations can be synthetically challenging. Here we report the intermolecular diamination of alkynes to afford  $\alpha$ -diimines from Ti imidos, alkynes, nitriles, and C-nitrosos. This proposed to occur by the [4+2]-cycloaddition of a nitroso to the Ti and carbon of a trapped [2+2] diazatitanacyclohexadiene intermediate, followed by two subsequent cycloreversion steps to eliminate nitrile and afford the  $\alpha$ -diimine and a Ti oxo.



# P47. Triptycenyl phosphines as platform for the design of frustrated Lewis pair

Mathieu Gama University of Namur, Namur, Belgium Graduate Student Advisor(s): Guillaume Berionni

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In 2008, Stephan defined FLPs as sterically hindered Lewis acids and bases that cannot form a covalent Lewis adduct because of steric repulsions.<sup>[1]</sup> In this work, the synthesis of triptycene-derived sterically hindered phosphines and their associations with reference Lewis acids systems of calibrated electrophilicities and Lewis acidities are investigated in the context of the development of new frustrated Lewis pairs.

1. D. W. Stephan, Org. Biomol. Chem. 2008, 6, 1535



# P48. Efficient Transfer Hydrogenation of Ketones using Methanol as Liquid Hydrogen Organic Carrier

<u>Nidhi Garg</u> Ph.D.- IIT Kanpur, Delhi, India Graduate Student **Advisor(s):** Dr. Basker Sundararaju

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Instead of conventional, stoichiometric reducing agents or hazardous  $H_2$  gas with transition metals, transfer hydrogenation of ketones with LOHC using transition metals is a viable alternative.But so far, only a few reports are there where methanol act as hydrogen source. We herein present, an efficient transfer hydrogenation of ketones by homogeneous iridium-bipyridonate system using methanol as LOHC. The methodology is efficient, practical, chemoselective with a wide array of ketones as scope.

Nielsen, M.; Alberico, E.; Baumann, W.; Drexler, H.-J.; Junge, H.; Gladiali, S.; Beller, M. *Nature* **2013**, 495, 85. Fujita, K.-I.; Kawahara, R.; Aikawa, T.; Yamaguchi, R. *Angew. Chem. Int. Ed.* **2015**, 54, 9057. Campos, J.; Sharninghausen, L. S.; Manas, M. G.; Crabtree, R. H. *Inorg. Chem.* **2015**, 54, 5079.



# P49. Unlocking the catalytic potential of tris(3,4,5-trifluorophenyl)borane using microwave irradiation

<u>Lukas Gierlichs</u> Cardiff University, Cardiff, United Kingdom Graduate Student **Advisor(s):** Rebecca Melen

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Enabling technology is rarely employed in main group catalysis. In this work microwave irradiation was used to enhance the reactivity of tris(3,4,5-trifluorophenyl)borane in the hydroboration of unsaturated substrates.

1. J. R. Lawson and R. L. Melen, *Inorg. Chem.*, **2017**, 56, 8627-8643.

2. J. L Carden, A. Dasgupta, and R. L. Melen, Chem. Soc. Rev., 2020, 49, 1706-1725.

3. Q. Yin, Y. Soltani, R. L. Melen and M. Oestreich, Organometallics, 2017, 36, 2381-2384.

4. C. O. Kappe, Angew. Chem. Int. Ed., 2004, 43, 6250-6284.

5. J. L Carden, L. J. Gierlichs, D. F. Wass, D. L. Browne, and R. L. Melen, *Chem. Comm.*, **2019**, 55, 318-321.Â



### P50. The Metalation of Neuronal Peptides with Cu(II)/(I) Modulates the Formation of Reactive Oxygen Species

<u>Camilla Golec</u> Trent University, Peterborough, Canada Undergraduate Student **Advisor(s):** Dr. Sanela Martic

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Tau protein aggregation has been linked to the pathogenesis of Alzheimer's disease (AD). However, the relationship between metal ions and the protein has not been characterized. Biologically relevant metal ions contribute to normal cellular function, but dyshomeostasis and the formation of reactive oxygen species (ROS) can lead to diseases<sup>(1-2)</sup>. Several spectroscopic methods were used to study the role of Cu(II)/(I) on the formation of ROS in the presence of brain-related tau peptides.

1. Faller, P.; Hureau, C. *Chem. Eur. J.* **2012**, 18, 15910. 2. Jiang, D.; Li, X.; Liu, L,; Yagnik, G.; Zhou, F.; Tabner, J.; Allsop, D. *J. Bio. Chem.* **2014**, 289, 12052.



# P51. Rational Design Approach for Improved Thermomechanical Properties in DPP-based Polymers

<u>Renée Goodman</u> Universoty of Windsor, Windsor, Canada Undergraduate Student **Advisor(s):** Dr. Simon Rondeau-Gagné

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Understanding the fundamental structure-mechanical property relationship for semiconducting polymers is essential for designing more flexible organic electronics and to develop rational tools for the preparation of soft conjugated polymers. Through backbone engineering, our team is investigating the effect of fused and isolated thiophene moieties incorporated to diketopyrrolopyrrole-based conjugated polymers, and the effect of side chains on the thermomechanical properties of the semiconductors.

S. Zhang, M. U. Ocheje, S. Luo, D. Ehlenberg, B. Appleby, D. Weller, D. Zhou, S. Rondeau-Gagné and X. Gu, *Macromol. Rapid Commun.*, **2018**, 39, 1-8
 S. Zhang, M. U. Ocheje, L. Huang, L. Galuska, Z. Cao, S. Luo, Y. H. Cheng, D. Ehlenberg, R. B. Goodman, D. Zhou, Y. Liu, Y. C. Chiu, J. D. Azoulay, S. Rondeau-Gagné and X. Gu, *Adv. Electron. Mater.*, **2019**, 5, 1-11.



#### P52. Introducing Hemilability to Cationic Indium Complexes

<u>Chatura Goonesinghe</u> University of British Columbia, Vancouver, Canada Graduate Student **Advisor(s):** Parisa Mehrkhodavandi

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Hemilabile ligands can influence the behavior of organometallic complexes. We report a series of cationic indium complexes bearing ligands with hemilabile pendant arms. Stability and temperature behavior of these complexes are correlated to the donicity pendant groups. The presence external donors result in fluxional behavior of the hemilabile arm with decoordination in the presence of strong donors. This behavior allows these ligands to be used for tuning reactivity cationic indium species.

 (1) Goonesinghe, C.; Roshandel, H.; Diaz, C.; Jung, H.-J.; Nyamayaro, K.; Ezhova, M.; Mehrkhodavandi, P. *Chem. Sci.* **2020**, 11 (25), 6485-6491.
 (2) Diaz, C.; Ebrahimi, T.; Mehrkhodavandi, P. *Chem. Commun.* **2019**, 55 (23), 3347-3350.
 (3) Jung, H.-J.; Chang, C.; Yu, I.; Aluthge, D. C.; Ebrahimi, T.; Mehrkhodavandi, P. *ChemCatChem* **2018**, 10 (15), 3219-3222.



# P53. High-Efficiency Blue Photoluminescence in the Cs<sub>2</sub>NalnCl<sub>6</sub>:Sb<sup>3+</sup> Double Perovskite Phosphor

<u>Matt Gray</u> The Ohio State University, Columbus, United States Graduate Student **Advisor(s):** Pat Woodward

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The luminescent properties of a lead-free double perovskite  $Cs_2NalnCl_6$  doped with  $Sb^{3+}$  are explored. The host crystal structure is a cubic double perovskite and has rock salt ordering of Na<sup>+</sup> and In<sup>3+</sup> ions. It is a wide bandgap compound (5.4 eV), and substitution with  $Sb^{3+}$  leads to strong absorption in the UV due to localized  $5s^2 \rightarrow 5s^15p^1$  transitions on  $Sb^{3+}$  centers. Radiative relaxation via a  $3P1 \rightarrow 1S0$  transition, leads to intense blue luminescence, centered at 445 nm, with a PLQY of 79%.

Gray, M. B.; Hariyani, S.; Strom, T. A.; Majher, J. D.; Brgoch, J.; Woodward, P. M. *J. Mater. Chem. C.*, **2020**, 8, 6797-6803



# P54. Elucidating the Mechanism of Oxidative Addition of Alkyl Substrates to [Ni(COD)(dppf)].

Megan Greaves University of Strathclyde, Glasgow, UK Graduate Student Advisor(s): David Nelson

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This work examines the kinetics of this reaction with alkyl substrates, and shows that Ni(dppf)<sub>2</sub> is the active nickel species in the reaction with alkyl halides.(1) These reactions are first order in Ni(COD)(dppf), dppf, and alkyl halide, and inverse first order in COD. The rate expression is consistent with a rapid formation of Ni(dppf)<sub>2</sub> and a subsequent irreversible reaction with alkyl halide. The experimental data is supported by DFT studies which suggest a halide abstraction mechanism.

Bajo, S.; Laidlaw, G.; Kennedy, A., R.; Sproules, S.; Nelson, D., J. Organometallics, **2017**, 36, 1662



# P55. It's Better with Salt: Aqueous Ring-Opening Metathesis Polymerization at Neutral pH

<u>Marcus Grocott</u> University of Birmingham, Birmingham, UK Undergraduate Student **Advisor(s):** Professor Rachel O'Reilly

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At present, acidic media is required to avert catalyst decomposition during Aqueous Ring-Opening Metathesis Polymerization, limiting its scope to non-pH-sensitive substrates. However, it is found that catalyst activity and lifetime can be improved by chloride salts at neutral pH. The ability to conduct aqueous metathesis at neutral pH is expected to enable new research avenues for polymer and organic synthesis by expanding the synthetic toolbox for reactions that involve pH-sensitive substrates.



# P56. Selective C-F bond Functionalization of Polyfluorocarbon Using Frustrated Lewis Pair Approach

<u>Richa Gupta</u> National University of Singapore, Singapore, Singapore Graduate Student **Advisor(s):** Dr. Rowan D Young

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We have developed a catalytic frustrated Lewis pair approach for selective C-F bond activation in polyfluoromethyl groups with phosphine bases. The resulting phosphonium salts can be functionalized via Wittig coupling, photoredox alkylation, nucleophilic transfer and hydrophosphination reactions to install diverse functional groups into the activated C-F position. This reaction allows simple access to fluoro compounds of current interest to pharmaceutical, agrochemistry, and materials sciences.

R. Gupta, D. Mandal, R. D. Young, *J. Am. Chem. Soc*, **2018**, 140, 10682. R. Gupta, D. Mandal, A. K. Jaiswal, R. D. Young, *J. Am. Chem. Soc.* **2020**, 142, 2572.



# P57. Exploring Magnetic XY Behavior in a Quasi-2D Anisotropic Triangular Lattice of Cu(II) by Functionalized Graphene

<u>Kriti Gupta</u> Indian Institute of Science Education and Research (IISER), India Graduate Student **Advisor(s):** Dr. Nirmalya Ballav

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Botallackite (Bo) (Cu<sub>2</sub>(OH)<sub>3</sub>Br), an anisotropic triangular spin lattice is stabilized over 2D reduced graphene oxide (rGO) via simple oxidation reduction reaction. Such an anchoring led to the oriented growth of Bo crystallites in Bo-rGO system. Bo-rGO was found to be magnetically active with a Néel transition at ~8.9 K, crossing over to possible XY anisotropy at ~5 K as revealed by dc and ac susceptibility measurements -an unprecedented observation in the field assigned to an interfacial effect.

Gupta, K.; Dadwal, A.; Jha, P. K.; Jain, A.; Yusuf, S. M.; Joy, P. A.; Ballav, N. *Inorg. Chem.* **2020**, 59, 6214.



P58. Hydrosilylative Reduction of Tertiary Amides to Amines Catalyzed by N-(Phosphinoaryl)anilido Complexes of Iron and Cobalt

<u>Dylan Hale</u> Dalhousie University, Halifax, Canada Graduate Student **Advisor(s):** Dr. Laura Turculet

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The synthesis and characterization of low-coordinate Fe(II) and Co(II) complexes supported by the monoanionic P,N-ligand N-(2-dicyclohexylphosphinophenyl)-2,6-diisopropylanilide are described. Three-coordinate (P,N)Fe-hexamethyldisilazide complex (2), and four-coordinate (P,N)Fe-(3-Fe) and (P,N)Co-alkyl (3-Co) complexes were evaluated as pre-catalysts for the hydrosilylative reduction of amides with PhSiH<sub>3</sub> (5 mol% pre-catalyst, 1 equiv. PhSiH<sub>3</sub>, 80°C, 1-24 h).



# P59. The Synthesis of Boron-Nitrogen Doped Indenofluorenes for Organic Electronic Applications

<u>Tyler Hannah</u> University of Calgary, Calgary, Canada Undergraduate Student **Advisor(s):** Warren Edward Piers

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This project is concerned with the synthesis of a boron-nitrogen doped indenofluorene for use as a non-fullerene acceptor in organic solar cell (OSC) applications. Previous work on this indenofluorene core was able to generate a power conversion efficiency of 2 % in an OSC device when used as an acceptor. Synthetic efforts focused on improving the device efficiency utilizing this B-N core. A synthetic pathway to the target was not able to be established, however a key intermediate was obtained.

1. Morgan, M. M.; Nazari, M.; Pickl, T.; Rautiainen, J. M.; Tuononen, H. M.; Piers, W.; Welch, G. C.; Gelfand, B. S. *Chem. Commun.* **2019**, 55, 11095-11098.



### P60. Synthesis and Coordination Properties of Antimony Based Pincer Ligands

<u>Alexander Harrison</u> University of Calgary, Calgary, Canada Graduate Student **Advisor(s):** Dr. Roland Roesler

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PCP-pincer complexes are among the most prominent representatives of their class. Comparatively, the chemistry of their heavier analogs has seen little development. Although the coordination chemistry of stibine ligands is well developed, no pincer complexes with antimony pendant donors have been reported.<sup>1</sup> Our investigation into these ligand platforms has yielded both surprising and expected reactivity with Rh and Ir. The synthesis and reactivity of this SbOCOSb pincer ligand will be presented.

1. Morales-Morales, D. Mini. Rev. Org. Chem. 2008, 5, 141-152.



#### P61. Characterization of Copper Hyponitrite Complexes using X-ray Absorption Spectroscopy

<u>Brenda Henriquez</u> Cal State Polytechnic University, Pomona, United States Graduate Student **Advisor(s):** Dr. Stieber

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Discrete diketiminate copper complexes formed in solution were analyzed by X-ray absorption spectroscopy. The X-ray absorption near edge structure analysis determined the oxidation state of compound 1, 2, and 4 to be +1, while compound 3 has an oxidation state of +2. The extended X-ray absorption determines coordination environment and determined compound 4 is consistent with a hyponitrite bound, suggesting one of the first examples hyponitrite complex was synthesized and isolated.



### P62. Germanium Catechol Complexes: Synthesis and Assessment of their Lewis Acidity and Catalytic Activity

<u>Andrew Henry</u> Western University, London, Canada Graduate Student **Advisor(s):** Kim M. Baines

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Main group catalysts are currently being explored as promising alternatives to transition metal-based catalysts. While there are multiple examples of silicon Lewis acids,<sup>[1,2]</sup> germanium Lewis acid catalysts remain scarce. The synthesis of bis(catecholato)germanium complexes with two nitrogen donors has recently been reported.<sup>[3]</sup> The synthesis of a variety of germanium catechol complexes, as well as an assessment of the Lewis acidity and catalytic activity will be presented.

 Maskey, R.; Schädler, M.; Legler, C.; Greb, L.; *Angew. Chem. Int. Ed.* **2018**, 57, 1717.
 Libermann-Martin, A. L.; Bergman, R. G.; Tilley, T. D.; *J. Am. Chem. Soc.* **2015**, 137, 5328.
 Glavinović, M.; Krause, M.; Yang, L.; McLeod, J. A.; Liu, L.; Baines, K. M.; Friščić, T.; Lumb, J. P.; *Sci. Adv.* **2017**, 3, e1700149.



## P63. CuO nanoparticles/Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> MXene hybrid Nanocomposites for Detection of Toluene Gas

Angga Hermawan

Institute of Multidisciplinary Research for Advanced Materials (IMRAM), Tohoku University, Sendai, Japan Graduate Student Advisor(s): Prof. Shu Yin

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We report a facile preparation of CuO/Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> MXene hybrids via electrostatic self-assembly. The CuO/Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> MXene exhibited the improved toluene gas sensing response (Rg/Ra) of 11.4, which is nearly 5 times higher than that of the pristine CuO nanoparticles (2.3) to 50 ppm of toluene at 250 °C. Such excellent performance showed the promising applications of metal oxides/2D hybrid materials for VOCs gas sensing.

Hermawan, A.; Zhang, B.; Taufik, A.; Asakura, Y.; Hasegawa, T.; Zhu, J.;, Shi, P.; Yin, S. *ACS Appl. Nano Mater.* **2020**, 3 (5), 4755-4766.



### P64. Di-tert-butyldiphosphatetrahedrane: Catalytic Synthesis and Reactivity of the Elusive Phosphaalkyne Dimer

<u>Gabriele Hierlmeier</u> University of Regensburg, Regensburg, Germany Graduate Student **Advisor(s):** Prof. Robert Wolf

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While tetrahedranes as a family are scarce, neutral heteroatomic species are all but unknown, with the only reported example being  $AsP_3$ .<sup>[1]</sup> We present the isolation of di-tertbutyldiphosphatetrahedrane, (tBuCP)<sub>2</sub>, a neutral heteroatomic molecular tetrahedron which also is the long-sought-after free phosphaalkyne dimer.<sup>[2]</sup> (tBuCP)<sub>2</sub> is formed from the monomer tBuCP in a nickel-catalyzed dimerization reaction using [(NHC)Ni(CO)<sub>3</sub>] (NHC = IMes, IPr).<sup>[3]</sup>

[1] B. M. Cossairt, M.-C. Diawara, C. C. Cummins, *Science* **2009**, 323, 602.

[2] A. Chirila, R. Wolf, J. C. Slootweg, K. Lammertsma, *Coord. Chem. Rev.* **2014**, 270-271, 57.

[3] G. Hierlmeier, P. Coburger, M. Bodensteiner, R. Wolf, *Angew. Chem. Int. Ed.* **2019**, 58, 16918; Angew. Chem. **2019**, 131, 17074.



#### P65. Iron Complexes with N-Heterocyclic Bidentate Carbene Ligands

<u>Alexis Hoxie</u> Cal State Polytechnic University, Pomona, USA Undergraduate Student **Advisor(s):** Dr. Chantal Stieber

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Iron complexes with bidentate N-Heterocyclic carbene ligands (RNHC<sub>2</sub>; R = alkyl or aryl) have not been widely reported and have potential to be catalysts for ranges of chemical transformations. This work aims to make new Fe<sup>0</sup> pre-catalysts with MesNHC<sub>2</sub> ligands. MesNHC<sub>2</sub> was synthesized from mesitylimidazole, followed by addition of CH<sub>2</sub>Br<sub>2</sub> to form a bis(imidazolium) salt. The salt was deprotonated with KHMDS and reacted with FeCl<sub>2</sub> to synthesize the corresponding FeCl<sub>2</sub> complex, (MesNHC<sub>2</sub>Me)FeCl<sub>2</sub>.



# P66. Tethered tungsten-alkylidenes for the synthesis of cyclic polynorbornene via REMP: Unprecedented stereoselectivity and trapping of key intermediates

<u>Vineet Jakhar</u> University of Florida, Gainesville, United States Graduate Student **Advisor(s):** Dr. Adam S. Veige

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This report describes an approach for preparing tethered tungsten-imido alkylidene complexes featuring a tetraanionic pincer ligand. Treating the tungsten alkylidyne [ $^{t}BuOCO$ ]W=C( $^{t}Bu$ )(THF)<sub>2</sub> with isocyanates leads to cycloaddition occurring exclusively at the C=N bond to generate the tethered tungsten-imido alkylidenes. Unanticipated intermediates reveal themselves, including the discovery of [( $O_2C^{t}BuC$ =)W((N,C)-RNCO)(THF)] and an unprecedented decarbonylation product [(tBuOCO)W(NR)(tBuCCO)].



### P67. Indium-catalyzed block copolymerization of lactide and methyl methacrylate by sequential addition

<u>Hyuk-Joon Jung</u> University of British Columbia, Vancouver, Canada Graduate Student **Advisor(s):** Prof. Parisa Mehrkhodavandi

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We synthesized a series of neutral and cationic indium complexes supported by tridentate Schiff base ligands and investigated their reactivities for the homo- and block copolymerization of lactide or ε-caprolactone and methyl methacrylate by sequential addition. The block copolymers were stringently characterized by size exclusion chromatography (SEC), diffusion ordered NMR spectroscopy (DOSY), differential scanning calorimetry (DSC), and fractional precipitation techniques.

Jung, H.-J.; Yu, I.; Nyamayaro, K.; Mehrkhodavandi, P. ACS Catal. 2020, 10, 6488-6496.



### P68. A multistimuli responsive heteroleptic iridium(III) complex: role of hydrogen bonding in probing solvent, pH and bovine serum albumin (BSA)

<u>Vishal Kachwal</u> (BITS Pilani, Pilani campus) Birla Institute of Technology, Pilani, India Graduate Student **Advisor(s):** Inamur Rahaman Laskar

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This article focuses on the vital role of hydrogen bonding to explain some unusual photophysical behaviors of an Aggregation-induced emission (AIE) active Iridium(III) complex. The preponderance of hydrogen bonding leads to the complex's multifunctional character, viz., sensing ability of base and protein (BSA), pH probing, and solvatochromism. The sensitivity of the complex to BSA was measured to be 9.3 pM.

Kachwal, V., Sharma, P. K., Sarmah, A., Chowdhury, S., & Laskar, I. R, *J. Mater. Chem. C*, **2020**, 8(19), 6605-6614.



#### P69. Ruthenium(II) Polypyridine Complexes for Photodynamic Therapy: From In Silico Prediction to In Vivo Application

<u>Johannes Karges</u> University of California, San Diego, San Diego, USA Post-doctoral Fellow **Advisor(s):** 

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The currently clinically applied photosensitizers are associated with many limitations. As an alternative, the use of Ru(II) polypyridine complexes from a theoretical DFT guided search to an in-depth photophysical and biological evaluation is presented. While being able to overcome all of the limitations, the lead compound demonstrated to eradicate a drug resistant tumour inside a mouse model upon clinically relevant 1-Photon or 2-Photon excitation.

 J. Karges, S. Kuang, F. Maschietto, O. Blacque, I. Ciofini, H. Chao, G. Gasser, ChemRxiv 2020, DOI: 10.26434/chemrxiv.11767995.v1
 J. Karges, S. Kuang, Y. C. Ong, H. Chao, G. Gasser, ChemRxiv, DOI: 10.26434/chemrxiv.12440012.v1.



#### P70. Nuclear Medicine at War: Fighting Cancer

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Jaspreet Kaur, Dhandeep Singh



#### P71. Magnesium Cobaltates: Synthesis and Reactivity

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Generally, d-block metallates contain alkene or polyarene ligands and alkali metal cations, mainly lithium and potassium, yet the influence of these countercations on reactivity has been poorly explored. Our recent research has focused on d-block metallates stabilised by group 2 cations. Herein, we present the synthesis and structural characteristics of 2, the first magnesium complexes of the [Co(cod)<sub>2</sub>]-anion. The reactivity and coordination chemistry of these species will be discussed.

[1] Ellis, J. E.; Dalton Trans., 2019, 48, 9538

[2] Jonas, K.; Mynott, R.; Krüger, C.; Sekutowski, J. C.; Tsay, Y-H.; *Angew. Chem. Int. Ed.*Engl., 1976, 15, 767. [3] (a) Hoidn, C. M.; Rodl, C.; McCrea-Hendrick, M. L.; Block, T.;
Pöttgen, R.; Ehlers, A. W.; Power, P. P.; Wolf, R.; *J. Am. Chem. Soc*, **2018**, 140, 13195. (b)
Büschelberger, P.; GÓ"rtner, D.; Reyes-Rodriguez, E.; Kreyenschmidt, F.; Koszinowski, K.;
Jacobi von Wangelin, A.; Wolf, R.; *Chem. Eur. J.*, **2017**, 23, 3139. (c) C. M. Hoidn, T. M.
Maier, K. Trabitsch, J. J. Weigand, R. Wolf, *Angew. Chem. Int. Ed.*, **2019**, 58, 18931. (d) C.
Rödl, R. Wolf, *Chem. Eur. J.*, **2019**, 8332.



### P72. Construction of photoresponsive coordination polymers for tuning electrical conductivity

<u>Samim Khan</u> Aliah University, Kolkata, India Post-doctoral Fellow **Advisor(s):** Dr. Md. Hedayetullah Mir

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Coordination polymers (CPs) have recently been used for the fabrication of electronic devices. A Cu(II) CP (1) has been synthesized which exhibits response with visible light enabling repetitive switching of the photocurrent displaying substantial on-off ratio. Secondly, a Cd(II) based CP (2) has been synthesized which undergoes [2+2] cycloaddition reaction to generate dimerized CP (3). Compound 2 shows better charge transport property than 3, may be due to  $\pi$ - $\pi$  stacking interactions in 2.

 Khan, S.; Halder, S.; Ray, P. P.; Herrero, S.; Gonzalez-Prieto, R.; Drew, M. G. B.; Chattopadhyay, S.; *Cryst. Growth Des.* **2018**, 18, 651.
 Islam, S.; Datta, J.; Maity, S.; Dutta, B.; Khan, S.; Ghosh, P.; Ray P. P.; Mir, M. H.; *Cryst. Growth Des.* **2019**, 19, 4057.



# P73. Three-in-one is really better: exploring the sensing and adsorption properties in a newly designed metal-organic system incorporating a copper (ii) ion

Mohammad Shahnawaz Khan Aligarh Muslim University, Aligarh, India Graduate Student Advisor(s): Dr Mohd Khalid

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In this work we have successfully designed a 2-D copper(II) based coordination polymer for the functional applications. Fluorescence sensing property demonstrates the high selectivity and sensitivity towards nitrobenzene and Fe<sup>3+</sup>. Moreover, CP (1) acted as tremendous adsorbent material for natural dyes with different charges and size i.e. methylene blue (MB), methyl orange (MO) and rhodamine-B.

[1] Khan, M. S.; Khalid, M.; Ahmad, M. S.; Shahid, M.; & Ahmad, M. *Dalton Trans.* **2019**, 48, 12918.

[2] Nagarkar, S. S.; Joarder, B.; Chaudhari, A. K.; Mukherjee, S.; & Ghosh, S. K. Angew. Chem. **2013**, 125, 2953.



#### P74. Bio-derived and Biodegradable Lignin-graft-Poly(lactic acid) Copolymers

<u>Diana Kim</u> University of British Columbia, Vancouver, Canada Graduate Student **Advisor(s):** Dr. Parisa Mehrkhodavandi

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Lignin is the second most abundant bio-derived material on earth after cellulose. Nonetheless, this cheap and renewable material has been underutilized due to the complexity of its structure. In this study, we aim to functionalize lignin with controlled molecular weight and enantiopure biodegradable poly(lactic acid) (PLA). The prepared PLA is successfully grafted to lignin via graft-to synthetic strategy and a series of lignin-graft-PLA copolymers are synthesized.

1. Novaes, E.; Kirst, M.; Chiang, V.; Winter-Sederoff, H.; Sederoff, R., *Plant Physiol.* **2010**, 154 (2), 555-561.

2. Iguchi, Y.; Akasaka, S.; Asai, S., Polym. J. 2020, 52 (2), 225-235.

3. Kadla, J. F.; Kubo, S.; Venditti, R. A.; Gilbert, R. D.; Compere, A. L.; Griffith, W.,. *Carbon* **2002**,40 (15), 2913-2920.

4. Johansson, C.; Bras, J.; Mondragon, I.; Nechita, P.; Plackett, D.; Simon, P.; Gregor Svetec, D.; Virtanen, S.; Giacinti Baschetti, M.; Breen, C.; Aucejo, S., *Bioresources* **2012**, 7 (2), 2506-2552.

5. Tsuji, H., Macromol. Biosci. 2005, 5 (7), 569-597.

6. Sun, Y.; Yang, L.; Lu, X.; He, C., J. Mater. Chem. A 2015, 3 (7), 3699-3709.

7. Chile, L.-E.; Kaser, S. J.; Hatzikiriakos, S. G.; Mehrkhodavandi, P., *ACS Sustainable Chem. Eng.* **2018**,6 (2), 1650-1661. 8. Yu, I.; Acosta-Ramirez, A.; Mehrkhodavandi, P., **2012**, 134 (30), 12758-12773.



### P75. Monoanionic pincer complexes of bismuth: potential new class of Lewis acid catalysts

<u>Marcus Kindervater</u> Dalhousie University, Halifax, Canada Graduate Student **Advisor(s):** Dr. Saurabh Chitnis

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Herein we debut the coordination chemistry of a rigid P<sub>2</sub>N<sub>3</sub> phosphinimine and a flexible PNP pincer complex of bismuth that show varying degrees of planarity. We further demonstrate the ability for PNP-Bi complexes to support highly electrophilic Bi(III) centers with potential application as Lewis acid catalysts.

Kindervater, M. B.; Hynes, T.; Marczenko, K. M.; Chitnis, S. S. *Dalton Trans.* **2020**. https://doi.org/10.1039/D0DT01413C.



### P76. Structural Transformations of a Mechanochemically Generated Polymorphic Organocalciate

Ross Koby Vanderbilt University, Nashville, United States Graduate Student Advisor(s): Timothy P. Hanusa

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Ball milling K[A'] (A' =  $[1,3-(SiMe_3)_2C_3H_3]$ -) with Cal<sub>2</sub> yields [KCaA'], which initially forms a structure (1) likely containing sigma-bound allyl ligands. Over time, the compound rearranges to a form (2) with eta 3-bound allyl ligands, which can be crystallized as a coordination polymer. The transformation from 1 to 2 can be followed by NMR and has been modeled with DFT calculations. Form 2 is capable of initiating polymerization of methyl methacrylate and isoprene at extremely low temperature.



### P77. The Rate of Cadmium Metallation of Metallothionein is Dependent on Structure and Binding Pathways

<u>Natalie Korkola</u> Western University, London, Canada Graduate Student **Advisor(s):** Dr. Martin Stillman

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The protein metallothionein (MT) binds up to 7 divalent metals using 20 cysteines.<sup>[1]</sup> Metallation proceeds by either binding to terminal cysteines or through bridging "clusters".<sup>[2]</sup> Here, the kinetics of these cadmium metallation pathways using stopped flow combined with UV-visible spectroscopy were studied. The beaded pathway proceeded much faster than the cluster pathway. Unfolding of apo-MT also slowed down the rate.<sup>[3]</sup> This provides a further understanding of MT's metallation pathways.

1. Sutherland, D. E. K.; Stillman, M. J., *Metallomics* **2011**, 3, 444-463.

- 2. Irvine, G. W.; Pinter, T. B.; Stillman, M. J., Metallomics 2016, 8 (1), 71-81.
- 3. Wong, D. L.; Korkola, N. C.; Stillman, M. J., *Metallomics* 2019, 11 (5), 894-905.



### P78. Combining Metal-Metal and Metal-Ligand Cooperativity Using a Naphthyridine-based Proton-responsive PNNP 'Expanded pincer'

<u>Errikos Kounalis</u> Utrecht University, Utrecht, The Netherlands Graduate Student **Advisor(s):** Dr. ing. D.L.J. (Danny) Broere

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Inspired by nature,<sup>[1-2]</sup> we have synthesized a proton responsive 'expanded pincer' ligand that can bind multiple 1st-row transition metals in close proximity and facilitate metal-metal and metal-ligand cooperativity. We have demonstrated cooperative activation of H<sub>2</sub> on a dicopper (I) complex, concomitant with the formation of a uniquely shaped Cu<sub>4</sub>H<sub>2</sub> cluster.<sup>[3]</sup> Additionally, organocopper complexes and their remarkable structural behavior will be discussed.<sup>[4]</sup>

Sippel, D.; Rohde, M.; Netzer, J.; Trncik, C.; Gies, J.; Grunau, K.; Djurdjevic, I.; Decamps, L.; Andrade, S.L.A, Einsle, O. *Science*. **2020**, 359, 6383, 1484-1489.
 Can, M.; Armstrong, F.A.; Ragsdale, S.W. *Chem. Rev.* **2014**, 114, 8, 4149-4174.
 Kounalis, E.; Lutz, M.; Broere, D.L.J. *Chem. Eur. J.* **2019**, 25, 58, 13280-13284.
 Kounalis, E.; Lutz, M.; Broere, D.L.J. *Organometallics* **2020**, 39, 4, 585-592.



### P79. Autonomous Chemical Robot Discovers the Rules of Coordination Chemistry

Daniel Kowalski University of Glasgow, Glasgow, UK Graduate Student Advisor(s): Prof Lee Cronin

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Digital chemistry is a rapidly growing discipline, although the adoption of automation has been slower in coordination chemistry than other areas.<sup>[1-3]</sup> We demonstrate the use of an unsupervised algorithm to robotically search a predefined chemical space for novel coordination complexes based on the data from in situ analysis.<sup>[4]</sup> Without any prior knowledge or simulation of chemistry, the algorithm has permitted the isolation of multiple new complexes and further in situ observations.

 [1] Coley, C. W.; Eyke, N. S.; Jensen, K. F. Angew. Chem. Int. Ed. **2020**, 59, ASAP
 [2] Gromski, P. S.; Henson, A.; Granda, J.; Cronin L. Nat. Rev. Chem. **2019**, 3, 119
 [3] Henson, A. B.; Gromski, P. S.; Cronin L. ACS Cent. Sci. **2018**, 4, 793
 [4] Porwol, L.; Kowalski, D. J.; Henson, A.; Long, D.-L.; Bell, N. L.; Cronin L. Angew. Chem. Int. Ed. **2020**, 59, 11256



#### P80. Fused Inorganic Rings Containing Germanium and Boron Towards Hybrid Materials

<u>Kelsie Krantz</u> University of Virginia, Charlottesville, United States Graduate Student **Advisor(s):** Robert J. Gilliard, Jr.

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The synthesis of highly conjugated organic molecules has advanced light emitting diodes and other electronic display technologies. By incorporating other p-block elements into organic systems, improved optoelectronic properties are achieved. This poster will highlight a new class of pyrene-fused N-heterocyclic germylenes which display extremely large twist angles. Additionally, the isolation, characterization, and photoluminescent properties of their boron analogues will also be discussed.

Krantz, K. E.; Weisflog, S. L.; Frey, N. C.; Yang, W.; Dickie, D. A.; Webster, C. E.; Gilliard, R. J. Jr., *Chem. Eur. J.* **2020**, DOI: 10.1002/chem.202002118 Krantz, K. E.; Weisflog, S. L.; Yang, W.; Dickie, D. A.; Frey, N. C.; Webster, C. E., Gilliard, R. J. Jr., *Chem. Commun.* **2019**, 55, 14954



### P81. Synthesis and Characterization of Heteroleptic Redox Active Bismuth Dithiolene and Dithiolate Complexes

<u>Viveka Kulkarni</u> Mount Allison University, Sackville NB, Canada Undergraduate Student **Advisor(s):** Dr. Glen Briand

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Redox catalysis often feature toxic transition metal centers that can be harmful to both human health and the environment. Bismuth is a potential alternative due to its relatively low toxicity and cost, but does not possess accessible oxidation states. The current study focuses on the use of redox non-innocent ligands to impart redox activity on a heteroleptic bismuth complexes. The synthesis and structural characterization of two bismuth 3,4-toluenedithiolate complexes is reported.



### P82. Dinuclear Reductive Elimination of $H_2$ from an Ir(III)-H is Made Possible by Metal/Ligand Proton Tautomerism

<u>Jonathan Kuo</u> University of Pennsylvania, Philadelphia, United States Post-doctoral Fellow **Advisor(s):** Karen Goldberg

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We have synthesized a square-planar Ir(I) carbonyl compound bearing a protic bispyrazolatopyridine ligand. Additional ligands PPh<sub>3</sub> or <sup>t</sup>BuPy induce tautomerism, where a ligand-based proton migrates to the metal forming an octahedral Ir(III)-H. In solution, the hydride undergoes the formal dinuclear reductive elimination of H<sub>2</sub>, generating the corresponding Ir(II) dimer. This is possible because the hydride establishes a [<sup>t</sup>BuPy] dependent equilibrium with its ligand protonated Ir(I) tautomer.



#### P83. Heteroatom-Rich Ligands for Heavy Metal Sequestration

<u>Jan-Willem Lamberink</u> Western University, London, Canada Graduate Student **Advisor(s):** Johanna Blacquiere, Paul Ragogna

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Contamination of generated effluent with heavy metals is an issue that plagues many industries. The phosphane-ene reaction can generate phosphorus rich networks that are promising materials for heavy metal sequestration. Modification and study of cross-linkers provides crucial insight into the effectiveness of the overall material. These studies have led to the use of an underused methodology for P-H bond addition to C=N bonds, which can be utilized in the synthesis of heteroatom-rich polymers.

- 1. Barakat, M. A. Arab. J. Chem. 2011, 4(4), 361.
- 2. Cuthbert, T. et al. Catal. Sci. Technol. 2017, 7(13), 2685
- 3. Buckler, S. A.; Epstein, M. Tetrahedron. 1962, 18(11), 1221



#### P84. Ligand-Assisted Volatilization and Thermal Stability in Mo(VI) Compounds

<u>Michael Land</u> Carleton University, Ottawa, Canada Graduate Student **Advisor(s):** Sean Barry

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Herein, we investigate the volatility and thermal stability of the parent bis(tertbutylimido)dichloromolybdenum(VI) compound, and the effect of neutral ligands including amines, phosphines, ethers, and an N-heterocyclic carbene. Their volatility and thermal stability are assessed using thermogravimetric analysis and differential scanning calorimetry, respectively<sup>[1]</sup>. The compounds have also been ranked using a Figure of Merit system<sup>[2]</sup> based upon observed thermal behavior.

[1] Land, M. A.; Robertson, K. N.; Barry, S. T. *Organometallics*, **2020**, 39, 916.
[2] Griffiths, M. B. E.; Dubrawski, Z. S.; Bacic, G.; Japahuge, A.; Masuda, J. D.; Zeng, T.; Barry, S. T. *Eur. J. Inorg. Chem.* **2019**, 15, 4927.



#### P85. Solubility Measurements of Strontium Carbonate in Aqueous Ammonium Chloride for Aqueous Extraction of Strontium Carbonate from Strontium Molybdates

Andres Lara Contreras

Department of Chemistry and Chemical Engineering - Royal Military College of Canada, Kingston, Canada Graduate Student **Advisor(s):** Dr. Emily Corcoran and Dr. Jennifer Scott

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Sr-Mo-O compounds are synthesized to understand the thermochemistry of nuclear fuels. The formation of SrCO<sub>3</sub> can affect subsequent thermodynamic measurements. A solid-liquid SrCO<sub>3</sub> extraction method has been developed using aqueous NH<sub>4</sub>Cl, which preserves the solid Sr-Mo-O phase intact. Given the lack of S-L equilibria data for the SrCO<sub>3</sub> + NH<sub>4</sub>Cl + H2O system, a preliminary assessment of solubility limits in the temperature range 298 to 348 K and NH<sub>4</sub>Cl concentration of 0.25 to 2.00 M is presented.



#### P86. Iron Coordination Complexes for Ultra-High Efficiency Monolayer Electrochromic Materials

Nadia Laschuk Ontario Tech University, Oshawa, Canada Graduate Student Advisor(s): Dr. Olena V. Zenkina

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Terpyridine (tpy) coordinated to Fe(II) forms a coloured complex. Electrochemical oxidation from Fe(II) to Fe(III) under low potentials results in a reversible bleaching of the complex, known as electrochromic materials (ECMs). We synthesized and characterized a series of iron(II) bis(tpy) complexes, and then covalently attached them to novel conductive inorganic nanoparticle (NP) supports. Together, these created atom efficient ECMs with ultrahigh durability and high colouration efficiencies.

Laschuk, N.I.; Obua, A.; Ebralidze, I.I.; Fruehwald, H.M.; Poisson, J.; Egan, J.G.; Gaspari, F.; Naumkin, F.Y.; Easton, E.B.; Zenkina, O.V. ACS *Appl. Electron. Mater.* **2019**, 1, 1705-1717.



### P87. Exploring Through-Bond and Through-Space Magnetic Communication in 1,3,2-Dithiazolyl Radical Complexes

Dominique Leckie University of Windsor, Windsor, Canada Graduate Student Advisor(s): Dr. Jeremy Rawson

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The field of single molecule magnets (SMM) has provided access to systems in which individual bits of data can be stored at the molecular (nm and sub-nm) scale. The use of paramagnetic ligands has attracted attention to increase the spin ground state in polynuclear complexes. While ligands comprising heavier p-block elements have the potential to increase magnetic anisotropy. Here, we present the characterization of prototypical complexes of the thiazyl radical MBDTA with M(hfac)<sub>2</sub>.<sup>(1)</sup>

(1) Leckie, D.; Stephaniuk, N. T.; Arauzo, A.; Campo, J.; Rawson, J. M., *Chem. Commun.* **2019**, 55 (66), 9849-9852.



### P88. 2-(2'-Pyridyl)-4,6-diphenylphosphinine: A Chemically Non-Innocent Ligand Toward Iron(0) and Nickel(0)

<u>Julia Leitl</u> Institute of Inorganic Chemistry, University of Regensburg, Regensburg, Germany Graduate Student **Advisor(s):** Prof. Dr. Robert Wolf

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Phosphinines are versatile ligands for transition metals due to their special steric and electronic properties.<sup>[1]</sup> In this work, new iron(0) complexes [Cp\*FeL]- (1- $\sigma$  and 1- $\pi$ , Cp\* = C<sub>5</sub>Me<sub>5</sub>) containing the chelating ligand 2-(2'-pyridyl)-4,6-diphenylphosphinine (L) have been prepared and found to undergo facile reaction with CO<sub>2</sub> under ambient conditions.<sup>[2]</sup> In contrast, L forms of a homoleptic phosphinine complex [Ni(L)<sub>2</sub>] (4) with a Ni(0) precursor, which shows reactivity toward phenyl halides.

1. P. L. Floch, Coord. *Chem. Rev.* **2006**, 250, 627-681. 2. J. Leitl, M. Marquardt, P. Coburger, D. J. Scott, V. Streitferdt, R. M. Gschwind, C. Müller, R. Wolf, *Angew. Chem. Int. Ed.* **2019**, 58, 15407-15411.



### P89. How Do Metal lons Affect the Ability of a Flavonoid to Quench the Superoxide Anion Radical?

Tyra Lewis

Department of Forensic Science, Environmental and Life Sciences, Trent University, Peterborough, Canada Undergraduate Student Advisor(s): Dr. Sanela Martic

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Quercetin (QCR) is a strong antioxidant against superoxide anion radical (SAR)<sup>(1)</sup>. While biologically-relevant metal ions regulate SAR, their roles on activity of QCR is not fully understood. By using electrochemical method<sup>(2)</sup>, the ability of Fe(III) and Cu(II) to modulate QCR activity was measured for metal ion:QCR ratio and complex type. Data indicated that antioxidant ability of QCR was regulated by metal ion, whilst the Cu(II)-QCR complex exhibited greatest antioxidant activity with SAR.

1. Gil, E.S.; Couto, R.O. Braz. J. Pharmacogn. 2013, 23 (3), 542.

2. Zabik, N.L.; Anwar, S.; Ziu, I.; Martic-Milne, S. Electrochim. Acta. 2019, 296, 174.



#### P90. Mesoionic N-heterocyclic olefin carbene tautomerization

<u>Qiuming Liang</u> University of Toronto, Toronto, Canada Graduate Student **Advisor(s):** Datong Song

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N-heterocyclic olefins (NHOs), sometimes referred to as deoxy-Breslow intermediates, have been widely studied and found broad applications in coordination chemistry and (organo)catalysis. We herein present an unprotected mesoionic N-heterocyclic olefins (mNHO) which can tautomerize into mesoionic carbenes (MIC) at room temperature. Experimental and computational studies allow us to compare their thermodynamic properties. An unusual O<sub>2</sub> activation will also be presented.



#### P91. Mechanism of 8-Aminoquinoline Directed Ni-Catalyzed C(sp<sup>3</sup>)-H Functionalization: Unanticipated Mononuclear and Dinuclear Paramagnetic Ni(II) Species.

<u>Jun Yang Liu</u> University of Windsor, Windsor, Canada Graduate Student **Advisor(s):** Dr. Samuel Johnson

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The catalytic activation of C-H bonds using first row transition metals has been an ideal choice in constructing new C-C bonds, but first row metal such as Ni, is less reactive/selective. A common strategy is using directing groups(8-AQ). Although a wide range of Ni-catalyzed C-H activations are reported, studies probing the mechanisms remain limited. This study aims at identifying catalytically-relevant key intermediates, and describing the mechanism of directed nickel-catalyzed C-H arylation.



#### P92. Polymeric pH-Activated Nanoparticles for Lipo-toxic Cell Application

Kaitlynn Lopes Boston University, Boston, United States Graduate Student Advisor(s):

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Non-alcoholic fatty acid liver disease (NAFLD) is characterized by cellular lipotoxicity, increased lysosomal pH, and inhibited autophagic processes. We propose the development of polymeric pH-activated nanoparticles for acid delivery to the lysosomes within lipo-toxic cells. Nanoparticles with an average diameter of approximately 100 nm and a polydispersity between 0.1 and 0.2 were deemed ideal for cellular uptake. Optimal nanoparticles were formed and determined to be non-cytotoxic.

 Younossi, Z., Anstee, Q.M., Marietti, M, Hardy, T., Henry, L., Eslam, M., George, J., & Bugianesi, E. *Nature Reviews Gastroenterology & Hepatology.* **2018**, 15,11-20.
 Trudeau, K.M., Colby, A.H., Zeng, J., Las, G., Feng, J.H., Grinstaff, M.W., Shirihai, O.S. *J Cell Biol.* **2016**, 214 (1), 25-34.
 Feldstein, A.E., Werneburg, N.W., Canbay, A., Gicciardi, M.E., Bronk, S.F., Rydzewski, R., Burgart, L.J., Gores, G.J. *AASLD.* **2004**, 40 (1), 185-194.

4. Turro; N.J. and Lei, X. Langmuir. 1995, I (1), 2525-2533.



#### **P93. Tellurium-Containing Polymers for Organic Electronics**

<u>Bruno Luppi</u> University of Alberta, Edmonton, Canada Graduate Student **Advisor(s):** Eric Rivard

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The field of polymer electronics often relies on complex molecular/polymeric structures and metastable device architectures to achieve high-performance.<sup>[1,2]</sup> This poster describes the synthesis of structurally simple Te-containing polymers of interest for photovoltaics.<sup>[3]</sup> These show reduced band gaps (e.g. 1.3 eV) when compared to their structurally analogous polythiophenes.<sup>[4]</sup> Additionally, cycloalkyl-fused oligomers reveal a drastic band gap reduction when the cycloalkyl ring size is altered.

1. L. Dou, Y. Liu, Z. Hong, G. Li and Y. Yang, *Chem. Rev.* **2015**, 115, 12633.

2. L. Lu, T. Zheng, Q. Wu, A. M. Schneider, D. Zhao and L. Yu, *Chem. Rev.*, **2015**, 115, 12666.

3. B. T. Luppi, D. Majak, M. Gupta, E. Rivard and K. Shankar, *J. Mater. Chem. A*, **2019**, 7, 2445.

4. B. T. Luppi, R. McDonald, M. J. Ferguson, L. Sang and E. Rivard, *Chem. Commun.*, **2019**, 55, 14218.



#### P94. Benzene Insertion into a Nickel Hydride Bond

<u>Casper Macaulay</u> Dalhousie University, Halifax, Canada Post-doctoral Fellow **Advisor(s):** Mark Stradiotto

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Treatment of (PN)NiNHDipp (PN = N-phosphinoamidinate ligand) with Me<sub>2</sub>PhSiH in benzene solvent yielded the crystallographically characterized, antifacial-coordinated, dinuclear species (PN)Ni( $\mu^2$ - $\eta^3$ :  $\eta^3$ -C<sub>6</sub>H<sub>8</sub>)Ni(PN) (1), representing the first isolated example of benzene insertion into a Ni-H bond. Efforts to generate (1) via  $\beta$ -hydride elimination of alkyl complexes instead afforded low-coordinate agostic alkyl complexes, including (PN)Ni(Neopentyl), the first neutral Ni(II)  $\gamma^3$ -agostic alkyl complex.

1. Macaulay, C. M.; Samolia, M.; Ferguson, M. J.; Sydora, O. L.; Ess, D. H.; Stradiotto, M.; and Turculet, L. *Dalton Trans.* **2020**, 49, 4811-4816.



### P95. Synthesis of sterically hindered 9-phosphatriptycenes and computational investigation of their properties

Damien Mahaut University of Namur, Namur, Belgium Graduate Student Advisor(s): Prof. G. Berionni

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Frustrated Lewis Pairs (FLPs) consist in bulky Lewis acid and base which cannot form a Lewis adduct because of steric repulsions and have been extensively used as transition metal-free catalysts for hydrogenations.<sup>[1]</sup> In this context, 9-phosphatriptycenes constitute promising Lewis bases due to their cage-like structures. In this poster, the synthesis of sterically hindered 9-phosphatriptycenes will be presented as well as a DFT investigation of their properties and potential for FLP chemistry.

1a. Stephan, D. W., *Science* 2016, 354, 1248.
1b. Lam, J.; Szkop, K. M.; Mosaferi, E.; Stephan, D. W.; *Chem. Soc. Rev.* 2019, 48, 3592.



#### P96. Photophysical and Optical Properties of Semiconducting Polymer Nanoparticles Prepared from Hyaluronic Acid and Polysorbate

<u>Aleena Malik</u> Univeristy of Windsor, Windsor, Canada Undergraduate Student **Advisor(s):** Dr. Simon Rondeau- Gagné

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Diketopyrrolopyrrole-based semiconducting polymer nanoparticles prepared from hyaluronic acid and polysorbate were prepared and characterized using transmission electron microscopy, dynamic light scattering, small-angle neutron scattering, and optical spectroscopy to determine their photophysical and optical properties. These nanoparticles were found to exhibit characteristics that are similar to the native polymers and show promise for application in biomedical imaging and nanomedicine.



### P97. S-S antioxidant assessment

Basel Mansour University of Windsor, Windsor, Canada Graduate Student Advisor(s): James W. Gauld

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The assessment of the antioxidant activity of different compounds contains S-S bond by using computational methods.



#### P98. Bismuthanyl-Stibanes: The first neutral Bi-Sb σ-bonds

<u>Kate Marczenko</u> Dalhousie University, Halifax, Canada Graduate Student **Advisor(s):** Dr. S. Chitnis

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Thermally-robust bismuthanylstibanes<sup>[1]</sup> are prepared in a one-step, high yield reaction, providing the first examples of neutral Bi-Sb  $\sigma$ -bonds in the solid state. DFT calculations indicate that the bis(silylamino)naphthalene scaffold is well-suited for supporting otherwise labile bonds. The reaction chemistry of the Bi-Sb bond is exploited by showing insertion of a sulfur atom, providing the first documented example of a Bi-S-Sb bonding moiety.

[1] Marczenko, K. M., Chitnis, S. S., Chem. Commun., 2020, In Press.



# P99. Dendronized Side Chains in Conjugated Polymers Toward More Robust Organic Electronics

<u>Gage Mason</u> University of Windsor, Windsor, Canada Graduate Student **Advisor(s):** Simon Rondeau-Gagne

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Organic semiconductors are used in organic electronics due to their solution processability and optoelectronic properties. Efforts have been made to improve the mechanical properties and solubility while enhancing stability. Several methodologies have been utilized to improve these properties, but sidechain engineering remains one of the most efficient. Recently, our group has incorporated dynamic bonding moieties (ligands or hydrogen bonding) to improve the thermomechanical properties.

- 1. Heeger, A. J. et al. Adv. Mater. 2014, 26, 2993-2998.
- 2. Hou, J. et al. Chem. Mater. 2014, 26, 3603-3605.
- 3. Bazan, G. C. et al. Angew. Chem., Int. Ed. 2014, 53, 14378-14381.
- 4. Brabec, C. J. et al. Adv. Funct. Mater. 2014, 24, 1449-1457.
- 5. Andreasen, J. W. et al. J. Mater. Chem. A 2015, 3, 17022-17031.
- 6. Tian, W. et al. Sol. Energy Mater. Sol. Cells 2012, 97, 28-33



# P100. Synthetic Efforts Towards Biologically Active Chelated Boron Complexes

<u>Kathleen May</u> Ryerson Univeristy, Toronto, Canada Graduate Student **Advisor(s):** R. A. Gossage and S. A. Westcott

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Boron-containing compounds have been overlooked when considering drug design. More recently popularized by the discovery of 5-fluorobenzoxaborole, boron-containing compounds have seen increased use and research interest. We intend to take advantage of the known biological activity of various azoles and use them with those a boron centre to increase biological efficacies. Ligand synthesis and advancements as well as future directions to determine their biological activity will be noted.



# P101. Nickel-Catalyzed Cross-Coupling of Sulfonamides With (Hetero)aryl Chlorides

<u>Ryan McGuire</u> Dalhousie University, Halifax, Canada Graduate Student **Advisor(s):** Dr. Mark STradiotto

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The development of Ni-catalyzed C-N cross-couplings of sulfonamides with (hetero)aryl chlorides is reported. These transformations, which were previously achievable only with Pd catalysis, are enabled by use of air-stable (L)NiCl(o-tol) pre-catalysts (L=PhPAd-DalPhos and PAd<sub>2</sub>-DalPhos), without photocatalysis. The collective scope of (pseudo)halide electrophiles demonstrated herein is unprecedented for any reported catalyst system for sulfonamide C-N cross-coupling.

McGuire, R. T.; Simon, C. M.; Yadav, A. A.; Ferguson, M. J.; Stradiotto, M. Angew. Chem. Int. Ed. **2020**, 59, 8952-8956.



### P102. Computational Studies on the Formation of Biological Per/Polysulfides

Paul Meister University of Windsor, Windsor, Canada Graduate Student Advisor(s): Prof. James W. Gauld

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Sulfur is a rare element in biology and typically exists as one of two proteinogenic amino acids, cysteine and methionine. They have important roles in redox signaling, protein folding, and enzymatic function. Persulfides (RSSH) are reactive intermediates making them a difficult topic to study. Herein, we used computational methods to study the formation of persulfide bonds and the extension of a sulfur chain within sulfide:quinone oxidoreductase (SQR) from a hyperthermoacidophilic archaea.



# P103. Probing the Sequence of Copper-Thiolate Cluster Formation in Metallothionein Using Native ESI-Mass Spectrometry

<u>Adyn Melenbacher</u> Western University, London Graduate Student **Advisor(s):** Professor Martin Stillman

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Cu is vital for life. Free cellular Cu catalyzes the formation of dangerous radicals. Cellular copper chaperones are required to avoid uncontrolled redox chemistry. Metallothionein (MT) binds Zn(II) and Cu(I). ESI-MS can be used to view the series of Cu(I)-thiolate clusters forming in MT, however it is unclear which cysteines bind to the Cu(I) first. Isolated domain fragments of MT have been used to determine domain preference of Cu(I) binding through detailed native ESI-MS studies.

Scheller, J. S.; Irvine, G. W.; Wong, D. L.; Hartwig, A.; Stillman, M. J. *Metallomics*. **2017**, 9, 447.



# P104. Mechanistic Investigation of the Nickel-Catalyzed Carbonylation of Alcohols

<u>Maximilian Menche</u> Catalysis Research Laboratory (CaRLa), Germany Graduate Student **Advisor(s):** 

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The carbonylation of alcohols represents an atom-efficient methodology for the preparation of carboxylic acids. Current processes either rely on harsh reaction conditions or precious metals<sup>[1-2]</sup>. Here, we present a mechanistic study of a catalytic system based on Nil<sub>2</sub>, which can regioselective carbonylate benzylic alcohols. Quantum-chemical computations were used to evaluate the underlying mechanistic processes. They revealed that two mechanisms are responsible for the observed reactivity.<sup>[3]</sup>

[1] Samel, U.; Kohler, W.; Gamer, A. O.; Keuser, U.; Yang, S.; Jin, Y.; Lin, M.; Wang, Z. Propionic acid and derivatives in Ullmann's Encyclopedia of Industrial chemistry, Wiley-VCH, Weinheim, **2014**.

[2] Elango, V.; Murphy, M. A.; Smith, B. L.; Davenport, K. G.; Mott, G. N.; Moss, G. L. EP0284310A1, **1988**.

[3] Sabater, S.; Menche, M.; Ghosh, T.; Krieg, S.; Rück, K. S. L.; Paciello, R.; Schäfer, A.; Comba, P.; Hashmi, A. S. K.; Schaub, T. *Organometallics* **2020**, 39, 870-880.



# P105. Exploring the photocatalytic hydrogen production by ruthenium polypyridine and cobaloxime assemblies

<u>AB MIR</u> IIT Gandhinagar, Baramulla, India Undergraduate Student **Advisor(s):** Dr Armab Dutta

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Axial coordinated imidazole derivatives; such as imidazole carboxylic acids and histidine; directly linked an identical Co(dimethylglyoxime)<sub>2</sub> core is shown to catalyze the reduction of aqueous proton to hydrogen (H<sub>2</sub>) in light-driven photocatalytic reactions in water near neutral pH.



P106. Sustainable Synthesis of Pyridines, Quinolines and Pyrimidines by Ru Catalyzed Acceptorless Dehydrogenation of Alcohols and C-C/C-N Coupling

<u>Rajarshi Mondal</u> University of Manitoba, Winnipeg, Canada Post-doctoral Fellow **Advisor(s):** Dr. David E. Herbert

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The acceptorless dehydrogenation of widely available biomass generated alcohols followed by coupling with amines can lead to sustainable and atom economical routes to core organic building blocks including nitrogen-containing heterocycles. In this presentation, preparation of a simple novel P^N ligand based Ru catalyst for acceptorless dehydrogenation of alcohols, substrate scope for N-heterocycle synthesis such as pyridines, quinolines and photo-emissive pyrimidines will be discussed.<sup>[1]</sup>

[1]. (a) Mondal, R.; Herbert, D. E. *Organometallics* **2020**, 39, 1310 (b) Mondal, R.; Lozada, I.B.; Davis, R. L.; Williams, J. A. G.; Herbert, D.E. *Inorg. Chem.* **2018**, 57, 4966 (c) Mondal, R.; Giesbrecht, P. K.; Herbert, D. E. *Polyhedron* **2016**, 108, 156.



### P107. Development of Green Solvent Processable Semiconductors for Printed Electronics

<u>Madison Mooney</u> University of Windsor, Windsor, Canada Graduate Student **Advisor(s):** Simon Rondeau-Gagne

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Our research aims to minimize the impact of electronic waste on the environment by developing a series of green solvent processable organic semiconductors for Printed Electronics. By incorporating hydrogen-bonding saccharide moieties through side-chain engineering, a series of semiconducting polymers were synthesized to be soluble in environmentally-friendly alcohol-based solvents, rather than the toxic halogenated solvents typically used for processing of these materials.



# P108. A fluorine transfer reaction: C-F bond activation leading to nucleophilic fluorination

Patrick Morgan University of Nottingham, Nottingham, United Kingdom Graduate Student Advisor(s): Pete Licence

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We present a first account of a transfer fluorination protocol where fluorine is transferred from a perfluorinated moiety onto a non-fluorinated substrate. The reaction occurred via the concurrent C-F bond activation, fluorine transfer, and C-F bond formation resulting in selective fluorination of an organic target. Fluorine transfer was tracked using in-situ ReactIR, where the rate of C-F bond activation was observed to match the rate of C-F bond formation.

(1) Morgan, P. J.; Hanson-Heine, M. W. D.; Thomas, H. P.; Saunders, G. C.; Marr, A. C.; Licence, P. *Organometallics* **2020**, 39 (11), 2116-2124.



### P109. The Influence of Carbon Donor Ligands on Biomimetic High-Spin Iron Complexes for N<sub>2</sub> Reduction

<u>Alexandra Nagelski</u> Yale University, New Haven, United States Graduate Student **Advisor(s):** Patrick Holland

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The active site clusters of nitrogenase enzymes have a C-based bridge between high-spin Fe sites. We present functional and structural models to elucidate the impacts of carbon and sulfide donors in simpler Fe compounds. This includes Fe alkyl and alkylidene complexes capable of reducing N<sub>2</sub> to NH<sub>3</sub> (with protons and electrons). We report the first diiron complex bridged by alkylidene and sulfide ligands, which models part of the active-site cluster (FeMoco) in the resting state of Mo-nitrogenase.



# P110. Halogen Bonding as a Structure-Directing Interaction in Dithiadiazolyl Radicals

<u>Mitchell Nascimento</u> University of Victoria, Victoria, Canada Graduate Student **Advisor(s):** 

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We report dithiadiazolyl radicals p-XC<sub>6</sub>F<sub>4</sub>CNSSN (X = Br (1) or I (2)). 1 is polymorphic, comprising pure diamagnetic  $\pi^*$ - $\pi^*$  dimers, fully monomeric radicals, and mixtures of each. 2 is isomorphous to 1 $\alpha$ . Both Br and I groups in 1 and 2 promote sigma-hole interactions of the type C···N (X = Br, I), reflecting the strength of this interaction in heavier halo-derivatives. A subtle competition of halogen/nitrogen and sulfur/nitrogen bonds directs structure and is examined through DFT methods.

1. Nascimento, M. A.; Heyer, E.; Less, R. J.; Pask, C. M.; Arauzo, A.; Campo, J.; Rawson, J. M. *Cryst. Growth Des.* **2020**, 20, 4313.



# P111. Olefin metathesis performance is impacted by X group reactivity in WOR<sub>3</sub>X/SiO<sub>2</sub> catalysts

<u>Chris Nicholas</u> Honeywell UOP, Des Plaines, United States Industry **Advisor(s):** 

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The novel complex  $W(=O)Np_3F$  has been synthesized and characterized by fluorination of the corresponding chloride counterpart with AgBF<sub>4</sub>. Reaction with SiO<sub>2</sub> proceeds both through silanolysis and opening of a siloxane bridge thanks to the affinity of silicon for fluoride. This material presenting fluorine on its surface shows an enhanced catalytic activity in propylene self-metathesis in comparison to its monopodal counterpart (SiO)W(=O)Np<sub>3</sub>.

1. Merle, N.; Mazoyer, E.; Szeto, K.C.; Rouge, P.; De Mallmann, A.; Berrier, E.; Delevoye, L.; Gauvin, R.M.; Nicholas, C.P.\*; Basset, J.-M.\*; Taoufik, M.\* *J. Organomet. Chem.* **2018**, 869, 11-17.

2. Merle, N.; Girard, G.; Popoff, N.; De Mallmann, A.; Bouhoute, Y.; TreÌbosc, J.; Berrier, E.; Paul, J.-F.; Nicholas, C.P.; Del Rosal, I.; Maron, L.\*; Gauvin, R.M.\*; Delevoye, L.\*; Taoufik, M.\* *Inorg. Chem.* **2013**, 52, 10119-30.

3. Mazoyer, E.; Merle, Nicolas; de Mallmann, A.; Basset, J.-M.\*; Berrier, E.; Delevoye, L.; Paul, J.-F.; Nicholas, C. P.; Gauvin, R. M.\*; Taoufik, M.\* *Chem. Commun.* **2010**, 46(47), 8944-8946.



### P112. Imparting Multifunctionality by Utilizing Biporosity in a Zr-Based MOF

<u>Pranay Ninawe</u> Indian Institute of Science Education and Research, Pune, India, Pune, India Undergraduate Student **Advisor(s):** Dr. Nirmalya Ballav

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An elegant design strategy of selectively incorporating conducting polymers in a bi-porous Metal-Organic Framework was introduced, which resulted in decoupled electrical and thermal conductivities. This process led to loading of ~12 wt% of polymers inside the pores. Electrical conductivity was enhanced by ~10^6 times in the nanocomposites bringing them into semiconducting regime, along with a significant decrease in thermal conductivity, as well as simultaneous porosity retention up to ~70%.

Jadhav, A.; Gupta, K.; Ninawe, P.; Ballav, N. Angew. Chem. Int. Ed. 2020, 59, 2215.



### P113. Activation of Ammonia and Hydrazine via the Formation of Terminal Fe(III) Amido and Imido Radical Intermediates

Lucie Nurdin University of Calgary, Calgary, Canada Graduate Student Advisor(s): Prof. Warren Piers

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Herein, we explore the use of a dianionic pentadentate ligand  $B_2Pz_4Py$  reported by our group  $^{[1,2]}$  for the activation of NH<sub>3</sub>. We hypothesized that this robust and electronically rich ligand could help to stabilize higher oxidation state complexes. In particular, we sought to detect terminal Fe(III)NH<sub>2</sub> and Fe(IV)NH complexes as they have been proposed as key intermediates of ammonia oxidation to N<sub>2</sub>, and could provide mechanistic insights to improve catalyst design using earth abundant metals.<sup>[3]</sup>

[1] Spasyuk, D. M.; Carpenter, S. H.; Kefalidis, C. E.; Piers, W. E.; Neidig, M. L.; Maron, L. *Chem. Sci.* **2016**, 7, 5939.

[2] Nurdin, L.; Spasyuk, D. M.; Fairburn, L.; Piers, W. E.; Maron, L. *J. Am. Chem. Soc.* **2018**, 140, 16094.

[3] Zott, M. D.; Garrido-Barros, P.; Peters, J. C. ACS Catal. 2019, 9, 10101.



### P114. Biodegradable electronics: Ionic diode based on a cellulose nanocrystals-agarose hydrogel

<u>Kudzanai Nyamayaro</u> University of British Columbia, Vancouver, Canada Graduate Student **Advisor(s):** Prof Mehrkhodavandi

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Herein, we employ cellulose nanocrystals to fabricated a polyelectrolyte gel diode capable of rectifying ionic current. The current rectification ratio reaches 70 reproducibly, which is significantly higher than analogous diodes generated with microfibrillated cellulose (~15) and olefin based polyelectrolyte gel diode (~40). The current voltage characteristics of the CNC-hydrogel diode were influenced by concentration, gel thickness, scanning frequency, and applied voltage.

1. Zhang, W.; Zhang, X.; Lu, C.; Wang, Y.; Deng, Y., *J. Phys. Chem. C* **2012**, 116 (16), 9227-9234.

2. Cayre, O. J.; Chang, S. T.; Velev, O. D., J. Am. Chem. Soc. 2007, 129 (35), 10801-10806.



# P115. Covalent Rigidification of Self-assembled Electroactive Materials with Polydiacetylenes

<u>Audithya Nyayachavadi</u> University of Windsor, Windsor, Canada Graduate Student **Advisor(s):** Simon Rondeau-Gagné

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Well-defined self-assembly of organic materials is an efficient strategy to control the optoelectronic properties of organic materials. However, supramolecular self-assembly are sometimes unstable and difficult to process. Here, we discuss our new approach to covalently stabilize self-assembled organic materials by incorporating diacetylenes capable of polymerizing in the solid state, preserving morphologies while tuning the optoelectronic properties of the resulting materials.

1. Tahir, M.N.; Nyayachavadi, A.; Morin, J.-F.; Rondeau-Gagné, S. *Polym. Chem.* **2018**, 9, 3019.

2. Nyayachavadi, A.; Langlois, A.; Tahir, M. N.; Billet, B.; Rondeau-Gagné, S. ACS Appl. Polym. Mater. **2019**, 1, 1918.

3. Tahir, M.N.; Abdulhamied, E.; Nyayachavadi, A.; Selivanova, M.; Eichhorn, S.H.; Rondeau-Gagné, S. *Langmuir.* **2019**, 47, 15158.



### P116. Computatinal Design of Fluorescent Chemosensors for Mercury Detection: Binding Affinities of Metal Ions with Thiacrown Ether

<u>Meagan Oakley</u> University of Minnesota, Minneapolis, United States Post-doctoral Fellow **Advisor(s):** 

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Design and improvement of fluorescence-based chemosensors for the detection of heavy metals remains a challenge. We have studied the BODIPY-benzothiacrown chemosensor<sup>[1]</sup> which showed selective chelation-enhanced fluorescence to the Hg<sup>2+</sup> metal ion. We computationally investigated the selectivity of the chemosensor as a positive fluorescent sensor for the Hg<sup>2+</sup> metal, as well as considering other crown ethers and other metal ions in order to understand what drives the selectiveness of the crown.

[1] Kim, H.-J.; Kim, S.-H.; Kim, J.-H.; Lee, E.-H.; Kim, K.-W.; Kim, J. S. *Bull. Korean Chem. Soc.* **2008**, 29, 1831-1834.



# P117. Pyrazine as Noncovalent Conformational Locks in Semiconducting Polymers for Enhanced Charge Transport and Stability in Thin Film Transistors

<u>Michael Ocheje</u> University of Windsor, Windsor, Canada Graduate Student **Advisor(s):** Simon Rondeau-Gagné

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A backbone engineering was performed on highly rigid semiconducting polymers based on diketopyrropyrrole (DPP) units to include pyrazine-containing moieties. Their utilization in bottom-gate bottom-contact devices and their potential to act as ligands for coordinating metals highlights the capacity of this new semiconductor for fully printed high-performance electronics.

Ocheje, MU; Goodman, RB; St. Onge, PBJ; Malik, MN; Yadiki, M; He, Y; Tao, Y; Chu, TY; Rondeau-Gagné, S *J. Mat. Chem. C*, **2019**, 7, 11507



#### P118. Synthesis of bis(imidazolium) salts with dimethylene benzene linkers

<u>Elisa Olivas</u> Cal State Polytechnic University, Bryn Mawr, United States Undergraduate Student **Advisor(s):** Dr. Chantal Stieber

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Bidentate N-heterocyclic carbene (NHC<sub>2</sub>) ligands are of interest as strongly sigma-donating ligands to metals. Previous work illustrates NHC<sub>2</sub> ligands with primarily ethylene and methylene linkers. This work focuses on synthesis of conjugated (bis)imidazolium salts by reacting mesitylimidazole with 1,2-di(bromomethyl) benzene or 1,2-dibromo-2-phenylethane in toluene. Characterization was with X-ray diffraction and <sup>1</sup>H NMR. The NHC<sub>2</sub> with 1,2-di(bromomethyl) benzene was successfully synthesized.

Hopkinson, M. N.; Richter, C.; Schedler, M.; Glorius, F. An Overview of N-Heterocyclic Carbenes. *Nature* **2014**, 510 (7506), 485-496



#### P119. Luminescent Difluoroboron Complexes Supported by Phenanthridine-Based β-Ketoiminates

<u>Robert Ortiz</u> University of Manitoba, Winnipeg, Canada Undergraduate Student **Advisor(s):** David Herbert

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Difluoroboron complexes have shown potential for developing cheaper materials for light emitting devices. Our group has developed synthetic methodologies that enable us to incorporate phenanthridine into various ligand frameworks, with their coordination complexes exhibiting interesting photophysical properties. Incorporating phenanthridine into  $\beta$ -ketoimines enabled us to isolate novel difluoroboron  $\beta$ -ketoiminates with photophysical properties dependent on the substituent on the phenanthridine.



### P120. Controlled Generation of 9-Boratriptycene by Lewis Adduct Dissociation: Accessing a Non-Planar Triarylborane

<u>Arnaud Osi</u> UNamur, Namur, Belgique Graduate Student **Advisor(s):** Prof. Guillaume Berionni

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The 9-boratriptycene was generated in solution by protodeboronation of a tetra-aryl boronate complex with HNTf<sub>2</sub>. We studied its stereoelectronic properties by quantifying its association with Lewis bases. Beyond the high pyramidalization of the 9-boratriptycene and its low reorganization energy upon coordination, theoretical calculations revealed that the absence of  $\pi$ -donation from the triptycene aryl rings to the boron vacant  $p_z$  orbital is the main reason for its high Lewis acidity.

Chardon, A.; Osi, A.; Mahaut, D.; Doan, T.-H.; Tumanov, N.; Wouters, J.; Fusaro, L.; Champagne, B.; Berionni, G. *Angew. Chem. Int. Ed.* **2020**, DOI: 10.1002/anie.202003119



#### P121. Boranes and Alanes: Novel Synthesis and Lactonisation

Darren Ould Cardiff University, Wales Graduate Student Advisor(s): Rebecca L. Melen

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Triaryl boranes have proven to be highly successful in catalysing organic reactions.<sup>[1,2]</sup> In this work the use of a series of triarylboranes to undergo metal-free lactonization of diazo compounds is presented, while also showing how tuning the Lewis acidity of the borane centre is important.<sup>[3]</sup> From here the synthesis and structural properties of a series of novel triarylalane species are discussed, highlighting key differences in the chemistry between triarylboranes and triarylalanes.<sup>[4]</sup>

[1] Lawson, J. R.; Melen, R. L. *Inorg. Chem.* **2017**, 56, 8627.
 [2] Jupp, A. R. and Stephan, D. W., *Trends Chem.*, **2019**, 1, 35.
 [3] Santi, M.; Ould, D. M. C.; Wenz, J.; Soltani, Y.; Melen, R. L.; Wirth, T. *Angew. Chem. Int. Ed.*, **2019**, 58, 7861.
 [4] Ould, D. M. C.; Carden, J. L.; Melen, R. L. *manuscript under revision*.



### P122. Convenient Access to Germanium and Aluminium Compounds Using the Pendant Power of Methyl-pyridinato Ligand

SANJUKTA PAHAR CSIR-NATIONAL CHEMICAL LABORATORY, Pune, India Graduate Student Advisor(s): Dr. Sakya Singha Sen

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A methyl-pyridine side arm in the nacnac, leads to a tridentate ligand (2,6-iPr<sub>2</sub>-C<sub>6</sub>H<sub>3</sub>NC(Me)CHC(Me)NH(CH<sub>2</sub>py)).We have synthesized and characterized chlorogermylene which undergoes unusual smooth ring contraction in presence of Lewis acid (GeCl<sub>2</sub>) via C-N bond cleavage, facile dehydrocoupling and 6 membered Al-heterocycle. With another Lewis acid, AlCl<sub>3</sub>, leads to the formation of dicholoaluminim complex via the transmetallation. Crystal reveals pyridine coordinates to the Al.

1. Xu, X.; Chen, Y.; Zou, G.; Sun, J. Single. *Dalton Trans.* **2010**, 39, 3952-3958 2. Pahar. S.; V.S.V.S.N. Swamy.; Das. T.; Gonnade. R. G.; Vanka. K.; Sen, S. S.; *Manuscript under preparation* 



### P123. Reactivity of Dioxygen and Nitric oxide towards Nonheme Diiron(II)hydrosulfide Complex

<u>Nabhendu Pal</u> Indian association for the cultivation of science, Kolkata, India Graduate Student **Advisor(s):** Dr. Amit majumdar

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While attempted synthesis of diiron(II)-hydrosulfide complexes using HS-produced insoluble precipitate, Reaction of Fe(BF<sub>4</sub>)<sub>2</sub>·6H<sub>2</sub>O, Et<sub>3</sub>N, and HN-Et-HPTB with RSH (R = PhCH<sub>2</sub>,tBu) /NaStBu in DMF at RT yielded the desired complex. The dioxygen activity of 1a to generate an oxygenated diiron complex at low temperature and its subsequent decomposition at RT has been examined. The reaction of 1a with 6 eq. Ph<sub>3</sub>CSNO produces [Fe<sub>2</sub>(*N*-Et-HPTB) (SH)(NO)<sub>2</sub>(DMF)](BF<sub>4</sub>)<sup>2</sup>.

Pal. N; Majumdar. A; Dalton Trans. 2019, 48, 5903-5908.



#### P124. High-throughput screening of MOF catalysts

<u>Joseph Palomba</u> UC San Diego, San Diego, United States of America Graduate Student **Advisor(s):** Seth Cohen

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Metal-organic frameworks (MOFs) are microporous hybrid materials whose metal and organic components make them useful heterogeneous catalysts for chemical transformations. MOFs combine the recyclability of heterogeneous materials with active site tailoring akin to homogeneous catalysts. We developed a high-throughput screening (HTS) method to examine >100 MOFs and MOF-polymer composites. This HTS method was tested on chemical warfare agent simulants to probe a broad spectrum of chemical space.



### P125. Rh(I) and Rh(III) pincer complexes: (R,R)-Ph-pybox vs iPr-PNP

<u>Gemma Parker</u> University of Warwick, Coventry, United Kingdom Graduate Student **Advisor(s):** Adrian Chaplin

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Pincer ligands are frequently used in organometallic chemistry and can be readily adapted, enabling the steric and electronic properties of metal derivatives to be tuned.<sup>(1)</sup> However, the impact of such changes can be difficult to gauge in a quantitative manner. To this end, we present our work investigating the rhodium(I) and rhodium(III) carbonyl fragments as probes for the donor properties of pincer ligands, exploiting convenient spectroscopic handles.<sup>(2)</sup>

 E. Peris and R. H. Crabtree, *Chem. Soc. Rev.*, **2018**, 47, 1959-1968.
 G. L. Parker, S. Lau, B. Leforestier and A. B. Chaplin, *Eur. J. Inorg. Chem.*, **2019**, 3791-3798.



#### P126. Heavy main-group carbonyl-alkyne metathesis

<u>Löwe Pawel</u> Westfälische Wilhelms-Universität Münster, Münster, Germany Graduate Student **Advisor(s):** Dr. Fabian Dielmann

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On this poster we show that oxophosphonium ions, which are the isoelectronic phosphorus congeners to carbonyl compounds, undergo [2+2] cycloaddition reactions with different terminal alkynes to generate cationic 1,2-oxaphosphetes. This new compound class was isolated and structurally characterized. The strained phosphorus-oxygen heterocycles open to the corresponding hetero-diene structure at elevated temperature, which was used to generate six-membered phosphorus heterocycles.

P. Löwe, M. Feldt, M. A. Wünsche, L. F. B. Wilm, F. Dielmann *J. Am. Chem. Soc.* **2020**, 142, 21, 9818-9826.



#### P127. How Does Milstein's Catalyst Really Work?

<u>John Pham</u> Colgate University, Hamilton, NY, United States Undergraduate Student **Advisor(s):** Anthony Chianese

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Mechanistic proposals of Milstein's ruthenium pincer catalyst for ester hydrogenation have relied on reversible deprotonation of a methylene linker of the pincer ligand. Here, we present evidence that under catalytic conditions, Milstein's catalyst undergoes ethane loss followed by hydrogen addition to give an active catalyst containing an N-H group. This newly discovered activated form is the most efficient catalyst for ester hydrogenation, giving over 10,000 turnovers at rt with no added base.



### P128. Conversion of amine from aromatic nitro compounds using Heterogeneous catalyst

Ariprasanth Ramalingam SRM University, Chennai, India Graduate Student Advisor(s): Prof.M.Sasidharan

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The reduction of aromatic nitro compound to amines is an important organic transformation. As amines are necessary for preparation of dye, drugs, herbicides, and pesticides and preparation of amines received great attention Here we demonstrate the synthesis of amines from nitro compounds using base metal catalysts. The catalysts were thoroughly investigated by a series of characterization techniques such as XRD, TEM, SEM, and XPS.



#### P129. Stabilization and significance of group 14 metallylenes

<u>Ravindra Raut</u> Indian Institute of Science Education and Research, Pune, Pune, India Graduate Student **Advisor(s):** Dr. Moumita Majumdar

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The heavier analogues of carbene are unique due to its one unoccupied and another lone pair orbital, which can mimic the transition metals. The stabilization of such challenging species either requires steric protection by bulky ligand or kinetic stabilization by extra donor group in ligand. Here, we have stabilized neutral and cationic forms of germylenes and stannylene in simple Schiff base multidentate ligands. The unusual reactivity is displayed in the current poster.



# P130. Cationic Alkyl-Indium Complexes: Tuning the Reactivity with Hemilability

<u>Hootan Roshandel</u> University of British Columbia, Vancouver, Canada Undergraduate Student **Advisor(s):** Prof. Parisa Mehrkhodavandi

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Although widely used to tune the chemistry of transition metals, there are only a handful of reported hemilabile systems that explore the reactivity of group-13 metals. Thus, based on previous work in our group, we developed a series of cationic alkyl-indium complexes containing a heterocyclic hemilabile side-arm, providing the first glance at using hemilability to significantly influence the reactivity of indium complexes towards the ring opening polymerization of epoxides and rac-lactide.

 Jung, H.-J.; Chang, C.; Yu, I.; Aluthge, D. C.; Ebrahimi, T.; Mehrkhodavandi, P., *ChemCatChem* **2018**, 10 (15), 3219-3222.
 Diaz, C.; Ebrahimi, T.; Mehrkhodavandi, P.. *Chem. Commun.* **2019**, 55 (23), 3347-3350.



# P131. Computational Insight into the Sulfilimine (S=N) Bond Formation in Collagen-IV.

<u>Anupom Roy</u> University of Windsor, Windsor, Canada Graduate Student **Advisor(s):** James W. Gauld

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Collagen IV is an essential component of the basement membrane in the tissues. It provides proper cellular structure by the formation of sulfilimine bond (S=N) between Methionine and Lysine or Hydroxylysine (cross-links) residues which can be formed with or without post-translational modification. The aim of the study is the investigation of the formation of sulfilimine bond and its nature in Collagen IV using multi-scale approach that included MD, QM-Cluster, QT-AIM, and NBO analysis.

1. Ronsein, G.E., Winterbourn, C.C., Di Mascio, P. and Kettle, A.J. *Free Radical Biology and Medicine*, **2014**, 70, 278-287.

2. Vanacore, R., Ham, A.J.L., Voehler, M., Sanders, C.R., Conrads, T.P., Veenstra, T.D., Sharpless, K.B., Dawson, P.E. and Hudson, B.G. *Science* **2009** 325, 1230-1234.



P132. Electrically Conductive and Photosensitive Heteronuclear Complexes: From Binuclear to Tetranuclear to Coordination Polymers.

<u>Sourav Roy</u> Indian Institute of Science, India Post-doctoral Fellow **Advisor(s):** 

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Recently inorganic-organic hybrids are found to be useful for the fabrication of active electronic devices.<sup>1-4</sup> Two Ni/Pb, one Co/Na and one Co/Hg/Na Schiff base complexes are prepared and their electrical properties are investigated which show significant on-off ratio when used as photo-switch. Unique supramolecular networks [C-H… $\pi$ (N3) and S…Br interactions] are observed in the complexes which can modulate the electrical conductivities.<sup>5</sup>

1) Dey, A.; Layek, A.; Roychowdhury, A.; Das, M.; Datta, J.; Middya, S.; Das, D.; Ray, P. P. *RSC Adv.* **2015**, 5, 36560.

2) Roy, S.; Halder, S.; Drew, M. G. B.; Ray, P. P.; Chattopadhyay, S. *New J. Chem.* **2018**, 42, 15295.

3) Roy, S.; Halder, S.; Drew, M. G. B.; Ray, P. P.; Chattopadhyay, S. ACS Omega **2018**, 3, 12788.

4) Roy, S.; Dey, A.; Ray, P. P.; Ortega-Castro, J.; Frontera, A.; Chattopadhyay, S. *Chem. Commun.* **2015**, 51, 12974.

5) Conesa-Egea, J.; Redondo, C. D.; Martinez, J. I.; Gomez-Garcia, C. J.; Castillo, O; Zamora, F.; Amo-Ochoa, P. *Inorg. Chem.* **2018**, 57, 7568.



### P133. Photocatalytic hydrogen evolution under red light using ruthenium(II) 2,4-di(pyridin-2-yl)-pyrimidine complexes as photosensitizers

<u>Mira Rupp</u> Université de Montréal, Universitaet Wuerzburg, Montréal, Canada Graduate Student **Advisor(s):** Garry Hanan; Dirk Kurth

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A series of non-symmetric 2,6-di(pyridin-2-yl)-pyrimidine ligands with peripheral pyridine substituents in different positions and their Ru(II) complexes are reported. They exhibit a red-shifted emission, longer excited-state lifetimes and higher luminescence quantum yields and are easier to reduce than analogous terpyridine (tpy) complexes. In H<sub>2</sub> evolution experiments under blue and red light, the complexes are more active but less stable than previously reported [Ru(tpy)2]<sup>2+</sup> photosensitizers.



#### P134. Synthesis and Reactivity of a Terminal Gallium Imide

<u>Artemis Saddington</u> University of Oxford, Oxford, United Kingdom Undergraduate Student **Advisor(s):** Simon Aldridge

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The potassium gallyl complex  $K_2[(NON)Ga]_2$  (NON = 4,5-bis(2,6-diisopropylanilido)-2,7ditertbutyl-9,9-dimethylxanthene) acts as a source of nucleophilic gallium. The reaction of  $K_2[(NON)Ga]_2$  with the bulky azide  $Ph_3SiN_3$  enables the isolation of a terminal gallium imide species. The imide contains a highly polarised Ga-N bond, enabling reaction with PhCCH, PhSiH<sub>3</sub>, PhCHO, CO<sub>2</sub>, N<sub>2</sub>O and CO.

1. Hicks, J.; Vasko, P.; Goicoechea, J. M.; Aldridge, S. *Nature* **2018**, 557, 92-95. 2. Heilmann, A.; Hicks, J.; Vasko, P.; Goicoechea, J.; Aldridge, S. *Angew. Chemie. Int. Ed.* **2020**, 132, 4927-4931.



### P135. Commodity Polymers to Functional Materials: Single Step Synthesis via Hydroaminoalkylation

<u>Sabrina Scott</u> University of British Columbia, Vancouver, Canada Graduate Student **Advisor(s):** Laurel Schafer

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In an effort to improve the accessibility of functional materials, vinyl terminated polypropylene was aminated via the atom-economic hydroaminoalkylation transformation in solvent free conditions. This work provides an inexpensive and resource conscious protocol to access mono-telechelic polymers with amine functionality, providing an alternative to the typical multistep syntheses often coupled with protection/deprotection procedures which fail to meet modern demands for "green" products.<sup>1-5</sup>

1. J. Y. Dong, Z. M. Wang, H. Hong and T. C. Chung, *Macromolecules*, **2002**, 35, 9352-9359. 2. S. Ji, T. R. Hoye and C. W. Macosko, *Polymer*, **2008**, 49, 5307-5313.

3. A. A. Nagarkar, A. Crochet, K. M. Fromm and A. F. M. Kilbinger, *Macromolecules*, **2012**, 45, 4447-4453.

4. S. Pal, F. Lucarini, A. Ruggi and A. F. M. Kilbinger, *J. Am. Chem. Soc*, **2018**, 140, 3181-3185.

5. M. Alizadeh and A. F. M. Kilbinger, *Macromolecules*, 2018, 51, 4363-4369.



#### P136. Electronic structure investigation of electron rich and deficient Cusalen complexes

<u>Anmol Sharma</u> University of the Fraser Valley, Abbotsford, Canada Undergraduate Student **Advisor(s):** Linus Chiang

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Metal complexes coordinated by salen ligands have attracted much attention due to their highly tunable nature and consequently diverse application in catalysis and medicinal purposes.<sup>[1-3]</sup> This poster will discuss the synthesis of three novel Cu-salen complexes and characterization via IR, UV-Vis, and NMR spectroscopies. The electronic structure of the target compounds, investigated by cyclic voltammetry, will also be discussed.

 Chiang, L., Herasymchuk, K., Thomas, F., Storr, T., *Inorg. Chem.*, **2015**, 54, 5970-5980
 Kanso, H., Clarke, R. M., Kochem, A., Arora, H., Philouze, C., Jarjayes, O., Storr, T., Thomas, F., *Inorg. Chem.*, **2020**, 59, 5133-5148
 Bellefeuille, D., Orio, M., Barra, A., Aukauloo, A., Journaux, Y., Philouze, C., Ottenwaelder, X., Thomas, F., *Inorg. Chem.*, **2015**, 54, 9013-9026



#### P137. Oligonuclear Ru(II) Assemblies Based on Ditopic Terpyridine Ligands

<u>Natali Shevchenko</u> Université de Montréal, Montréal, Canada Undergraduate Student **Advisor(s):** Garry S. Hanan

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Linear polytopic terpyridine ligands can form rod-type assemblies when complexed with metal ions like ruthenium and iron. In this study, ditopic terpyridine ligands are synthesized, complexed and characterized. Tertbutyl groups are introduced to the ligands to increase the solubility of long supramolecular assemblies in organic solvents. The complex's absorption and redox potentials are studied and compared to an iron complex of the monotopic terpyridine ligand with tertbutyl substituents.

- 1. T. Auvray, R. Sahoo, D. Deschenes, G. S. Hanan, *Dalton Trans.*, **2019**, 48, 15136-15143
- 2. Wang, J.; Hanan, G. S. Synlett, 2005,36 (41), 1252-1254

3. D. J. Diaz, S. Bernhard, G. D. Storrier, H. D. Abruna, *J. Phys. Chem. B*, **2001**, 105, 8746-8754.

4. M. Rupp, M. Sc. thesis, University of Wuerzburg, 2018

- 5. Laramée-Milette, B. M.Sc. Thesis; Université de Montréal, 2013
- 6. Nagy, E.M.; Pettenuzzo, A.; et al. Chem. Eur. J., 2012, 18(45), 14464-14472
- 7. B. Laramée-Milette, G. S. Hanan, Chem. Commun., 2017, 53, 10496-10499



#### P138. Terphenyl(bisamino)phosphines as Ligands in Gold-Catalysis

<u>Jan-Erik Siewert</u> LIKAT Rostock, Rostock, Germany Undergraduate Student **Advisor(s):** Christian Hering-Junghans

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Terphenyl(bisamino)phosphines have been identified as effective ligands in cationic gold(I) complexes for the hydroamination of acetylenes. These systems are related to Buchwald phosphines and their steric properties have been evaluated. Effective hydroamination was noted even at low catalyst loadings and a series of cationic gold(I) complexes has been structurally characterized clearly indicating stabilizing effects through gold-arene interactions.

1. D. S. Surry, S. L. Buchwald, Angew. Chem., Int. Ed. 2008, 47, 6331.

2. D. Malhotra, M. S. Mashuta, G. B. Hammond, B. Xu, Angew. Chem., Int. Ed. 2014, 53, 4456.

3. T. Witteler, H. Darmandeh, P. Mehlmann, F. Dielmann, Organometallics 2018, 108, 3064.

4. S. Ullrich, B. Kovaä• eviä‡, X. Xie, J. Sundermeyer, *Angew. Chem. Int. Ed.* **2019**, 58, 1033.

5. W. Petz, F. Weller, J. Uddin, G. Frenking, Organometallics 1999, 18, 619.

6. R. Dorta, E. D. Stevens, N. M. Scott, C. Costabile, L. Cavallo, C. D. Hoff, S. P. Nolan, *J. Am. Chem. Soc.* **2005**, 127, 2485.

7. C. A. Tolman, *Chem. Rev.* **1977**, 77, 313.

8. L. Falivene, R. Credendino, A. Poater, L. Serra, R. Oliva, V. Scarano, L. Cavallo, *Organometallics* **2016**, 35, 2286.



### P139. Cytotoxic activity studies against non-tumorigenic HEK293 and cancer HEPG2 cell lines by Vanadium (V) compounds

<u>Aniela Silva Nolasco</u> Universidad Juarez Autonoma de Tabasco, Tabasco, México Graduate Student **Advisor(s):** Irma Sanchez Lombardo

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Recent studies have established the ability of decavanadates that contain amine-based ligands and/or cations to exert different biological effects in vitro and in vivo.<sup>1-3</sup> In this study the cytotoxic activity studies against non-tumorigenic HEK293 human embryonic kidney cells and human liver cancer HEPG2 cells (IC 50 values) were performed for sodium decavanadate, metforminium decavanadate, sodium metavanadate and metformin hydrochloride in a concentration range of 10-100 µM.

1. Rehder, D. Future Med. Chem. 2012, 4, 1823-1837.

2. Aureliano, M. World J. Biol. Chem. 2011, 2, 215-225.

3. Sigel, A; Freisinger, E; Roland K. O. Metal Ions in Life *Sciences*. Berlin, Germany, **2019**; Vol.19. www.mils-WdG.com <u>https://doi.org/10.1515/9783110527872-008</u>.

4. Sanchez, I. et al. Eur J. Inorg. Chem. 2014, 27, 4581.



#### P140. Analysis of Solid-State Geometries and Ground-State Energies of Tin-Containing Complexes via DFT

<u>Kaitlyn Silverthorne</u> Ryerson University, Toronto, Canada Undergraduate Student **Advisor(s):** Dr. Daniel Foucher

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Polystannanes are of interest due to their low band-gap and semi-conductive properties. Several monomeric precursors have been synthesized and characterized via NMR (<sup>119</sup>Sn) spectroscopy and single-crystal XRD. My project focuses on analyzing these tin-containing small molecules via DFT to assess the accuracy of various functionals in predicting solid-state geometries and ground-state energies of both the open and hypercoordinate conformations of these complexes.

1. Pau, J.; D'Amaral, G. M.; Lough, A. J.; Wylie, R. S.; Foucher, D. A. *Chem. Eur. J.* **2018**, 24, 18762-18771



#### P141. Iron catalyzed hydroboration of alkenes and alkynes

<u>Arpita Singh</u> Texas Tech University, Lubbock, United States Graduate Student **Advisor(s):** Dr. Michael Findlater

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Iron complexes [BIAN]Fe(I)(toluene) and [BIAN]FeCl<sub>2</sub> (BIAN=bis(2,6-

diisopropylaniline)acenaphthene) have been employed in the selective reduction of alkynes and alkenes via hydroboration. The hydroboration of alkynes and alkenes in the presence of HBpin and an activator at 70 °C afforded linear vinyl and alkyl boronic esters, respectively. Selectivity up to 98% was observed for alkyl boronic esters and exclusive formation of trans product was observed for vinyl boronic esters.



#### P142. Small Molecule Activation by Transition Metal Polyhydride Clusters

<u>Kathleen Snook</u> Boston University, Boston, United States Undergraduate Student **Advisor(s):** John Caradonna

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Polyhydride transition metal clusters introduce a new type of catalyst with multiple reactive metal surfaces, reservoirs of electron density, and fluxional hydride ligands. We are investigating the topology of an open butterfly cluster  $[M_4(\mu_3-X)_2(\mu-X)_5X_4]^{n-}$  along with the hexatungsten clusters  $[W_6(\mu_3-Cl)_8Cl_6]^{2-}$  and  $[W_6(\mu-Cl)_{12}Cl_6]^{2-}$ . Successful syntheses of these clusters are confirmed by Raman and <sup>183</sup>W NMR spectroscopy, and SCXD studies.

- 1. Energy Environ. Sci. 2010, 3 (7), 884-890.
- 2. Global Biogeochemical Cycles. 2019, 33, 318-342.
- 3. Energy Fuels. 2009, 23 (4), 1874-1887.
- 4. Catal. Sci. Technol. 2016, 6 (12), 4048-4058.
- 5. J. Am. Chem. Soc. 1982, 104 (21), 5630-5633.

6. Inorganic Syntheses; Girolami, G.S., Sattelberger, A.P., Eds.; John Wiley & Sons, Inc., **2014**; Vol. 36, pp 19-23.

7. J. Am. Chem. Soc. 1995, 117 (31), 8139-8150.



#### P143. A Responsive-at-Metal Cyclometalated Platinum Luminophore

<u>Miguel Soto</u> University of British Columbia, Vancouver, Canada Post-doctoral Fellow **Advisor(s):** Mark MacLachlan

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Molecular emitters are very valued as they are used in a range of applications and technologies. When designing an emitter, there are three characteristics that are highly desirable to control: emitted color, emission decay, and quantum yield, which are adjusted through the selection of a metal core, and by the covalent modification of a ligand. Here, I will discuss how simple solvent molecules can get the job done by modulating non-covalent interactions in cyclometalated Pt(II) complexes

Soto, M. A.; Carta, V.; Andrews, R. J.; Chaudhry, M. T.; MacLachlan, M. J., *Angew. Chem. Int. Ed.* **2020**, 59, 10348



## P144. Tuning the Redox Properties of MnI-NHC Complex Through Remote Coordination Approach

<u>Anandi Srinivasan</u> Laboratoire d'Electrochimie Moléculaire LEM, Paris, Paris, France Post-doctoral Fellow **Advisor(s):** Dr. Orestes Rivada Wheelghan

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Modulating the electronics and reactivity of transition metal complexes by incorporating alkali metal substituents offers a powerful strategy for enhancing the performance electrocatalysts. Along these lines, we present the correlation of remote interactions of different alkali cations with the novel MnI complex, 1, and its effects on the redox behavior of Mn center. Additionally, the electrochemical behavior of 1 towards CO<sub>2</sub>, including the effects of added alkali salts will be presented



#### P145. Optimizing Semiconducting Materials Through M-L Interactions.

<u>Blake St. Onge</u> University of Windsor, Windsor, Canada Graduate Student **Advisor(s):** Simon Rondeau-Gagne

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Metal-Ligand (M-L) interactions in materials chemistry which involve the incorporation of sidechain engineering in semiconducting  $\pi$ -conjugated polymers have become an increasingly popular field of study. Our work concludes that utilizing semiconducting  $\pi$ -conjugated polymers with pincer ligands in the sidechains can incorporate Fe(II) centers which demonstrate an increase in device performance and mobilities.

P. B. St.Onge, T. Chen, A. Langlois, A. Younus, I. Jo Hai, B. Lin, Y. Chiu and S. Rondeau-Gagné, *J. Mater. Chem. C.* **2020**, DOI: 10.1039/D0TC01536A.



#### P146. Synthesis of Redox Active Organoindium Dithiolate Catalysts

<u>Alexander Stöckli</u> Mount Allison University, Sackville, NB, Canada Undergraduate Student **Advisor(s):** Glen Briand

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Traditional redox catalysts use precious transition metals and tend to produce toxic waste. Use of relatively non-toxic indium as a metal centre in these compounds would help alleviate this problem. However, indium has limited redox activity on its and so the use of a redox active ligand is needed. This work seeks to synthesize redox active indium compounds with varied physical properties. The synthesis and structural characterization of a novel indium 3,4-tolueneditholate complex is reported.



#### P147. C-C Bond Formation Reactions Using B(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub>

<u>Katarina Stefkova</u> Cardiff University, Cardiff, United Kingdom Graduate Student **Advisor(s):** Dr. Rebecca Melen

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The poster demonstrates the recent developments in  $B(C_6F_5)_3$  catalysed alkenylations as well as it shows the utilization of  $B(C_6F_5)_3$  with PMes<sub>3</sub> as Frustrated Lewis Pairs in higly stereoselective C-C bond formation to obtain E- alkenes from readily available styrenes.

1. Soltani, Y.; Dasgupta, A.; Stefkova, K. et al., CPRS, 2020,

2. Dasgupta, A.; Stefkova, K. et al., DOI: 10.1002/anie.202007176.



# P148. C(sp<sup>2</sup>)-C(sp<sup>2</sup>) homo-dimerization of a pincer scaffold: A synergistic experimental and computational study supports binuclear Ni(III)-Ni(III) mechanisms

<u>S. Maryamdokht Taimoory Taimoory</u> University of Michigan, Chemistry Department, Ann Arbor, USA Post-doctoral Fellow **Advisor(s):** Prof. Melanie S. Sanford

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Chemical transformations promoted by bimetallic systems are abundant in nature and catalytic processes.<sup>1,2</sup> This presentation will highlight our ongoing studies on of the key role of bimetallic Ni(III)-Ni(III) complexes in mediating the C(sp<sup>2</sup>)-C(sp<sup>2</sup>) reductive elimination on a catalytically relevant, quinoline-benzamide scaffold.

Lindahl, et. al. J. Inorg. Biochem. 2012, 106, 172.
 a) Diao et. al. J. Am. Chem. Soc. 2016, 138, 4779; b) Diao et. al. Angew. Chem. Int. Ed. 2016, 55,7534; c) Uyeda et. al. ACS Catal. 2017, 7, 936; d) Ritter et. al. J. Am. Chem. Soc. 2012, 134, 12002.



### P149. Ni-salen complexes bearing pendant phenol groups in the secondary coordination sphere

<u>Kaeden Teindl</u> University of the Fraser Valley, Abbotsford, Canada Undergraduate Student **Advisor(s):** Linus Chiang

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The secondary coordination sphere (SCS) often plays an essential role in enzymatic processes,<sup>[1]</sup> which has inspired efforts to integrate SCS design into synthetic targets.<sup>[2-4]</sup> To that end, this poster will discuss the synthesis and spectroscopic characterization of novel nickel-salen complexes with phenol moieties appended in their SCS.<sup>[5]</sup> Their electrochemical behaviors were probed by CV, and the subsequent one-electron oxidized species were prepared and investigated spectroscopically.

1. Jeoung, J.-H.; Dobbek, H. Science 2007, 318, 1461-1464

2. Sun, P; Yang, D; Li, Y; Wang, B; Qu, J. Dalton Trans. 2020, Advance Article

3. Hong, D; Kawanishi, T; Tsukakoshi, Y.; Kotani, H.; Ishizuka, T.; Kojima, T. *J. Am. Chem. Soc.* **2019**, 141, 20309-20317

4. Wijeratne, G.B.; Bhadra, M.; Siegler, M.A.; Karlin, K.D. *J. Am. Chem. Soc.* **2019**, 141, 17962-17967

5. Chiang et al. Chem. Eur. J. 2012, 18, 14117 - 14127



### P150. Structural Landscape of Lanthanide coordinated Anderson-Evans polyoxometalate

<u>Shailabh Tewari</u> Indian Institute of Technology, Delhi, New Delhi, India Graduate Student **Advisor(s):** Arunachalam Ramanan

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The struggle in rationalizing crystallization, a supramolecular reaction, for the targeted design of functional materials is to recognize the underlined possible pathways. The task is significantly challenging due to the obscurity of well-defined links between synthesis, structure and property. However, Polyoxometalates (POM), the intermediate soluble molecular analogues of the bulk oxides, may provide some insights.

1. Tewari, S. et al. Front. Chem. 2019, 6, 635.



#### P151. Gas phase characterization of a dinitrogen analogue of Zeise's salt

<u>Gilian Thomas</u> University of Victoria, Victoria, Canada Graduate Student **Advisor(s):** Prof. Scott McIndoe

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We discovered a derivative of Zeise's salt ( $[PtCl_3(C_2H_4)]^-$ ), the  $[PtCl_3(N_2)]^-$  complex spontaneously forms in the nitrogen-rich atmosphere of the mass spectrometer. Ion mobility separation of  $[PtCl_3(C_2H_4)]^-$ ,  $[PtCl_3(N_2)]^-$  and  $[PtCl_3(CO)]^-$ , and MS/MS experiments analyzed ligand binding strength for the isobaric complexes. These were in agreement with calculated values. Binding strength may allow  $[PtCl_3(N_2)]^-$  to be isolated with the potential for dinitrogen functionalization and catalysis in water.



#### P152. Impact of Sterics and Metal Choice on Isostructural Group VI Alkyne Metathesis Catalysts and the Selective Formation of Isomeric Metallacyclic Intermediates.

<u>Richard Thompson</u> Louisiana State University, Baton Rouge, United States Post-doctoral Fellow **Advisor(s):** Semin Lee

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Metallacyclobutadienes as part of a cycloaddition based mechanism for alkyne metathesis is an established paradigm with alternate species viewed as decomposition products. Recently we reported the exclusive formation of a dynamic metallatetrahedrane capable of metathesis raising question about the mechanism. Consequentially, we report the synthesis a library of both molybdenum and tungsten alkylidynes supported by sterically distinct ligands and the role metal identity has on intermediates.

Thompson, R. R.; Rotella, M. E.; Du, P.; Zhou, X.; Fronczek, F. R.; Kumar, R.; Gutierrez, O.; Lee, S., *Organometallics* **2019**, 38, 4054-4059.



### P153. Trans oxidative addition of arylfluorides to nickel N-heterocyclic carbene complexes

<u>Adrian Torres</u> Cal State Polytechnic University, Pomona, Pomona, United States Undergraduate Student **Advisor(s):** Dr. S. Chantal E. Stieber

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Nickel N-heterocyclic carbene (NHC) complexes with isopropyl groups were reported to yield trans addition of aryl fluorides. In this work, NHC ligands with methyl groups were studied for oxidative addition of aryl fluorides at nickel. The aryl fluoride nickel complex was obtained from an oxidative addition of hexafluorobenzene to the nickel(0) precursor. Complexes were characterized using NMR spectroscopy and X-ray crystallography. A trans addition product was observed for the methyl complex.

1) Schaub, T.; Backes, M.; Radius, U. Square-Planar (Pentafluorophenyl)Nickel(II) Complexes by Derivatization of a C-F Activation Product. *Eur. J. Inorg. Chem.* **2008**, 2008 (17), 2680-2690. https://doi.org/10.1002/ejic.200800213.



## P154. Clays modified with $[Fe(o-phen)_3]^{2+}$ complex for the removal of triclosan from aqueous solutions

Carlos Torres-Méndez

Escuela de Quimica, Facultad de Ciencias Quimicas y Farmacia, Universidad de San Carlos de Guatemala, Guatemala, Guatemala Undergraduate Student Advisor(s): PhD. Silvia Echeverrìa

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Bentonite (Bent) and vermiculite (Vrm) were modified by ionic exchange with [Fe(ophen)<sub>3</sub>]SO<sub>4</sub> complex. The modified clays showed a change of color to an intense red, characteristic of [Fe(o-phen)<sub>3</sub>]<sup>2+</sup> complex. Materials were characterized by ATR FTIR, the hybrid clays showed bands at 720, 848, 1428 cm<sup>-1</sup> assigned to bending vibrations of C-H bonds in the aromatic rings of the o-phen ligand. Comparative analysis on isotherms and the sorption capacity of Bent, Verm and hybrid clays is presented.

(1) Lei, C.; Hu, Y.; He, M. *Chem. Eng. J.* **2013**, 219, 361-370.
 (2) Kabadagi, A.; Chikkamath, S.; Manjanna, J.; Kobayashi, S. *Appl. Clay Sci.* **2018**, 165, 148-154.
 (3) Foo, K. Y.; Hameed, B. H. *Chem. Eng. J.* **2010**, 156 (1), 2-10.
 (4) Chen, X. *Information* **2015**, 6 (1), 14-22.



#### P155. Application of FLP Chemistry to Reactions with Bromoalkynes

<u>Jonah Toth</u> University of Toronto, Toronto, Canada Undergraduate Student **Advisor(s):** Douglas W. Stephan

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It was previously reported that sterically encumbered or electron deficient phosphines do not react with (bromoethynyl)benzene.<sup>[1]</sup> Our work has involved a systematic study of the reactivity induced by Lewis acids of varied strength to promote reaction between haloalkynes of the form RCCBr and a range of phosphines bearing either bulky or electron-withdrawing substituents. This study represents the first known application of FLP chemistry involving bromoalkyne substrates.

[1] Veinot, A.J.; Todd, A.D.K.; Robertson, K.N.; Masuda, J.D. Can. J. Chem. 2017, 96, 8-17.



### P156. Development of Cu-pyridinedicarboxamide Complexes as C-H Bond Activation Reagents

<u>Warren VandeVen</u> University of the Fraser Valley, Abbotsford, Canada Undergraduate Student **Advisor(s):** Linus Chiang

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Nature utilizes high valent metal centers in enzymes to functionalize relatively inert C-H bonds.<sup>1-2</sup> Synthetic Cu(III) complexes, ligated by a dianionic, tridentate pyridinedicarboxamide (PCA) ligand, have demonstrated success in such reactions.<sup>3-4</sup> This poster will discuss the synthesis and characterization of a Cu complex ligated by an electron-rich PCA ligand. A more electron donating ligand is expected to stabilize the Cu(III) metal center, improving C-H bond activation reactivity.

1. Nam, W. Acc. Chem. Res. 2007, 40, 522.

2. Latham, J.; Brandenburger, E.; Shepherd, S. A.; Menon, B. R. K.; Micklefield, J. *Chem. Rev.* **2018**, 118, 232.

3. Donoghue, P. J.; Tehranchi, J.; Cramer, C. J.; Sarangi, R.; Solomon, E. I.; Tolman, W. B. *J. Am. Chem. Soc.* **2011**, 133, 17602.

4. Bower, J.K., Cypcar, A.D., Henriquez, B., Chantal, S., Stieber, E., Zhang, S., *J. Am. Chem. Soc.* **2020**, 142, 8514.



### P157. The path from darkness to light: octahedral iridium(III) complexes to combat cancer.

<u>Gloria Vigueras</u> University of Murcia, Murcia, Spain Graduate Student **Advisor(s):** José Ruiz López (@jruizum)

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We will present our recent results on the synthesis and biological evaluation of different biscyclometalated Ir(III) complexes.<sup>[1-3]</sup> Modifying the cyclometalated ligand and/or ancillary ligand we can tuned their antitumoral activity. Thus, some of them are very active in several cancer cell lines in the dark while others, are able to activate by irradiation with visible light.

 Yellol, J.; Pérez, S. A.; Yellol, G.; Zajac, J.; Donaire, A.; Vigueras, G.; Novohradsky, V.; Janiak, C.; Brabec, V.; Ruiz, J. *Chem. Commun.* **2016**, 52, 14165.
 Pracharova, J.; Vigueras, G.; Novohradsky, V.; Cutillas, N.; Janiak, C.; Kostrhunova, H.; Kasparkova, J.; Ruiz, J.; Brabec, V. *Chemistry* **2018**, 24, 4607.
 Novohradsky, V.; Vigueras, G.; Pracharova, J.; Cutillas, N.; Janiak, C.; Kostrhunova, H.; Brabec, V.; Ruiz, J.; Kasparkova, J. *Inorg. Chem.* Front. **2019**, 6, 2500.



### P158. Investigations into the polymerization mechanism and reactivity of 1-phosphabutadienes

<u>Henry Walsgrove</u> University of British Columbia, Vancouver, Canada Graduate Student **Advisor(s):** Derek Gates

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We recently reported the polymerization of 1-phosphaisoprene by anionic initiation.<sup>[1]</sup> Studies of the polymer microstructure showed predominantly 3,4-addition across the C=C double bond, forming a polymer containing pendant phosphaalkene groups. Initiation studies using <sup>n</sup>BuLi show a 4,1-enchainment, in contrast to the proposed polymer microstructure resulting from 4,3-addition. Further investigation of the reactivity of 1-phosphabutadiene species will also be presented.<sup>[2,3]</sup>

1. Dück, K.; Rawe, B. W.; Scott, M. R.; Gates, D. P. *Angew. Chem. Int. Ed.* **2017**, 56, 9507. 2. Ohtsuki, K.; Walsgrove, H. T. G.; Hayahsi, Y.; Kawauchi, S.; Patrick, B. O.; Gates, D. P.; Ito, S. *Chem. Commun.* **2020**, 56, 774.

3. Walsgrove, H. T. G.; Dabringhaus, P.; Patrick, B. O.; Gates, D. P. *Can. J. Chem.* **2020**, 98, 373.



#### P159. Supramolecular Capsules from Iso-Chalcogenazole N-Oxides

<u>JinZhao Wang</u> McMaster University, Hamilton, Canada Graduate Student **Advisor(s):** Dr.Ignacio Vargas-Baca

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Amongst the many molecules that are capable of forming chalcogen bonds, iso-tellurazole Noxides stand out because of their ability to reversibly form macrocyclic aggregates that are persistent in solution<sup>[1,2]</sup>. In recent investigations, we have found a procedure that builds a dodecamer that folds onto itself forming a cavitand that encapsulates certain molecules such as cyclohexane, tetrahydrofuran and dioxane.

1. P.C. Ho, P. Szydlowski, J. Sinclair, P. J. W. Elder, J. Kübel, C. Gendy, L. Myongwon Lee, H. Jenkins, J. F. Britten, D. R. Morim, I. Vargas-Baca. *Nature* Commun. **2016**, 7:11299. 2. P.C. Ho, J. Rafique, L. M. Lee, H. A. Jenkins, J. F. Britten, A. L. Braga, I. Vargas-Baca.\* *Dalton Trans.*, **2017**, 46, 6570-6579.



P160. Oxidative addition reactions of tetrathiocins to group 10 metals: From structural diversity to sensor development.

Lara Watanabe University of Windsor, Windsor, Canada Graduate Student Advisor(s): Dr. Jeremy Rawson & Dr. Charles Macdonald

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Metal dithiolene complexes are ideal for various materials applications. Oxidative addition reactions of bis(benzo)-1,2,5,6-tetrathiocins to zero valent group 10 metal complexes leads to mono-, di- and hexa-nuclear complexes depending on the steric effects of the phosphine. We have exploited this synthetic methodology to provide a simple two-step route to functionalized dithiolate complexes that we can explore as sensors for Na<sup>+</sup> cations.

1. Dithiolene chemistry: *Synthesis*, properties, and applications, ed. E. I. Stiefel, Interscience, Hoboken, New Jersey, **2003**.

2. Wrixon, J.; Hayward, J.; Raza, O.; and Rawson, J. M., *Dalton Trans.* **2014**, 43, 2134. 3. (a) Wrixon, J.; Hayward, J.; and Rawson, J. M., *Inorg. Chem.*, **2015**, 54, 9384-9386; (b) Wrixon, J.; Ahmed, Z. S.; Anwar, M. U.; Beldjoudi, Y.; Hamidouche, N.; Hayward J. J.; and Rawson, J. M., *Polyhedron*, **2016**, 108, 115-121.

4. Watanabe, L. K.; Wrixon, J. D.; Ahmed, Z. S.; Hayward, J. J.; Abbasi, P.; Pilkington, M.; Macdonald, C. L. B., and Rawson, J. M., *Dalton Trans.*, **2020**, DOI: 10.1039/D0DT01678K.



### P161. Synthesis, characterization and cytotoxicity studies of multinuclear benzimidazole-based cyclometallated and cationic ruthenium(II) complexes

<u>Athi Welsh</u> University of Cape Town, Cape Town, South Africa Graduate Student **Advisor(s):** Assoc. Prof. Gregory S. Smith

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In this study, a series of trimeric benzimidazole ligands, and their respective trinuclear Ru(II) complexes were synthesized and screened for their anticancer activity against two breast cancer cell lines. Overall, the compounds show anticancer activity either comparable to or greater than that of cisplatin. Importantly, the compounds have relatively low toxicity towards the non-tumorigenic cells and were noted to be significantly more selective for breast cancer cells compared to cisplatin.

A. Welsh, L. Rylands, V. B. Arion, S. Prince and G. S. Smith, *Dalton Trans.*, **2020**, 49, 1143 - 1156.



### P162. Examination of the Effects of Solvents on Homogeneous CO<sub>2</sub> Reduction Electrocatalysis

<u>Lucienna Wolf</u> DePaul University, Chicago, United States Undergraduate Student **Advisor(s):** Kyle A. Grice

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Significant effort has been focused on designing more efficient catalysts for CO<sub>2</sub> reduction, yet there is minimal information regarding the effects of solvent environments on these systems. Electrochemical and spectroscopic data collected from experiments using the canonical fac-Re(bpy)(CO)<sub>3</sub>Cl electrocatalyst in various solvents were compared to the solvents' physical properties and their economic and environmental properties to determine the most optimal conditions for CO<sub>2</sub> reduction systems.

1. Blunden, J., D. S. Arndt, and G. Hartfield, Eds., **2018**: State of the Climate in 2017. *Bull. Amer. Meteor. Soc.*, 99 (8), Si-S310, doi:10.1175/2018

2. Lu, Q.; Jiao, F., Electrochemical CO<sub>2</sub> reduction: Electrocatalyst, reaction mechanism, and process engineering. **2016**; Vol. 29.

3. Grice, K. A.; Saucedo, C., Inorganic Chemistry 2016, 55, 6240-6246.

4. Bron, M., Infrared Spectroelectrochemistry. In Encyclopedia of Applied Electrochemistry, Kreysa, G.; Ota, K.-i.; Savinell, R. F., Eds. Springer. New York: New York, NY, **2014**; pp 1071-1075.

5. Gutmann, V. (1978) The donor-acceptor aproach to molecular interactions. New York: Plenum Press. 6. Dean, J. A.; Lange, N. A., Lange's handbook of chemistry. 15th ed. ed.; McGraw-Hill: New York, **1999**.



#### P163. Mechanistic Studies into Iron-Catalysed Alkene Isomerisation

<u>Callum Woof</u> University of Bath, Bath, United Kingdom Graduate Student **Advisor(s):** Dr Ruth L Webster

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We have investigated the catalytic isomerisation of alkenes using a  $\beta$ -diketiminate iron(II) complex and a hydride source. Whilst we initially hypothesised this to be an entirely redoxneutral process in line with similar transformations,<sup>[1,2]</sup> upon studying the reaction further with a broad range of techniques including synthetic, EPR and electrochemical studies we now believe that the active species is an Fe(I) complex forming in situ, and the reaction proceeds by a redox-active pathway.

[1] M. Espinal-Viguri, R. L. Webster et.al., *Chem. Eur. J.*, **2016**, 22, 11605
[2] M. Espinal-Viguri, R. L. Webster et.al., *J. Am. Chem. Soc*, **2019**, 141, 572



### P164. Rational design and synthesis of near-infrared phthalocyanine-based absorbing

<u>Aymen Yaseen</u> Gebze Technical University, Kocaeli, Turkey Graduate Student **Advisor(s):** Ayse Gürek, Fabienne Dumoulin

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Photosensitizers for photodynamic therapy must ideally be excited by NIR wavelengths to reach deep-seated tumors.<sup>(1)</sup> Phthalocyanines are very suitable as their max. absorption can be tuned from 680 to 800 nm.<sup>(2,3)</sup> To benefit from the tumor-targeting Enhanced Permeation and Retention effect,<sup>(4,5)</sup> organosilica nanoparticles are aimed to be prepared from phthalocyanines absorbing at 800 nm. A careful design and synthetic strategy of appropriately functionalized phthalocyanines will be presented.

(1) Mallidi, S.; Anbil, S.; Bulin, A. L.; Obaid, G.; Ichikawa, M.; Hasan, T. Theranostics **2016**, 6 (13), 2458-2487.

(2) Lo, P. C.; Rodriguez-Morgade, M. S.; Pandey, R. K.; Ng, D. K. P.; Torres, T.; Dumoulin, F. *Chem. Soc. Rev.* **2020**, 49 (4), 1041-1056.

(3) Topal, S. Z.; Kumru, U.; Atilla, D.; Gürek, A. G.; Hirel, C.; Tommasino, J.-B.; Luneau, D.; Berber, S.; Dumoulin, F.; Ahsen, V. Dalt. Trans. **2014**, 43 (18), 6897-6908

(4) Maeda, H.; Nakamura, H.; Fang, J. Adv. Drug Deliv. Rev. 2013, 65 (1), 71-79.

(5) Maeda, H.; Tsukigawa, K.; Fang, J. Microcirculation **2016**, 23 (3), 173-182.



### P165. Synthesis of Nickel Nitrosyl Complexes with Bidentate N-heterocyclic Carbene Ligands

<u>Zijie Zhang</u> Cal State Polytechnic University, Pomona, Pomona, United States Undergraduate Student **Advisor(s):** Dr. S. Chantal E. Stieber

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Activation of small molecules like NO offer insight towards biological reduction mechanisms. Nickel nitrosyl complexes were synthesized to examine the effect of NHC ligands on the nickel-NO bond. A new metal complex (MesNHC<sub>2</sub>Me)Ni(COD) was synthesized and structurally characterized. A new Ni-NO complex [(MesNHC<sub>2</sub>Me)Ni(NO)][BF4] was synthesized and characterized by IR spectroscopy, which is a new example of a three-coordinate nickel nitrosyl complex with strongly sigma donating ligands.

1. Nolan, S. N-Heterocyclic Carbenes: Effective Tools for Organometallic *Synthesis*. **2014**. 2. Hermann, W.A. et al. *J. Organomet.Chem.* **1999**, 572, 239.

3. Liu, J.; Chen, J.; Zhao, J.; Zhao, Y.; Li, L.; Zhang, H. *Synthesis*, **2003**, 17, 2661-2666. 4. Cowie, M.; Wells, K. D.; Ferguson, M. J.; Mcdonald, R. *Organometallics*, **2008**, 27(4), 691-703.

5. Yamamoto, C. D.; Zhang, Z.; Stieber, S. C. E. *Acta Crystallogr. Sect. E Crystallogr. Commun.* **2018**, 74 (10), 1396-1399.

6. Wright, A. M.; Zaman, H. T.; Wu, G.; Hayton, T. W. *Inorg. Chem.* **2014**, 53 (6), 3108-3116 7. Tskhovrebov, A. G.; Solari, E.; Scopelliti, R.; Severin, K. *Inorg. Chem.* **2013**, 52 (20), 11688-11690

8. Chattopadhyay, T.; Podder, N.; Banu, K. S.; Banerjee, A.; Ghosh, M.; Suresh, E.; Nethaji, M.; Das, D. *J. Mol. Struct.* **2007**, 839 (1-3), 69-75.

9. Golchoubian, H.; Kocheksaraee, S. K.; Moayyedi, G. *Polyhedron* **2017**, 123 (2), 376-384. 10. Meidine, M. F.; Pombeiro, A. J. L.; Nixon, J. F. **1999**, 839, 3041-3045.



#### P166. Organotin(IV) Complexes as Catalyst for Biodiesel Formation: Synthesis, Structural Elucidation and Computational Studies

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Here we have presented the synthesis of three novel triorganotin(IV) complexes. Spectroscopically characterized by NMR in solution and by FT-IR, X-ray diffraction analysis in solid state. The computational study was also performed. The homogeneous catalytic performance was evaluated for the transesterification of corn oil with methanol into biodiesel. The catalytic results revealed that the complexes are potential candidates for the development of new catalytic system for biodiesel production.

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